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GOVERNMENT OF MAHARASHTRA WATER RESOURCES DEPARTMENT

HYDROLOGY PROJECT (SW) HYDROLOGY PROJECT DIVISION, AURANGABAD



Polluted river water kills as many people as a nuclear explosion

NOW YOU CAN MAKE A DIFFERENCE.



WATER QUALITY LAB LEVEL-II, AURANGABAD

ANNUAL REPORT

YEAR 2011

EXECUTIVE ENGINEER

HYDROLOGY PROJECT DIVISION, AURANGABAD

CERTIFICATE OF COMPLIANCE



HYDROLOGY PROJECT DIVISION WATER QUALITY LAB LEVEL II - AURANGABAD

Near Hedgewar Hospital, Garkheda, Aurangabad – 431 005, Maharashtra, India.

has been assessed by International Certification Services Pvt. Ltd. and registered as complying with the requirements of the following International Standard:

ISO 9001:2008

The Quality Management System applicable to:

Scope: Collection, Testing And Analysis Of Water And Waste Water Samples.



Further clarifications regarding the scope of this certificate and the applicability of ISO 9001:2008 requirements may be obtained by consulting the above certification body.

International Certification Services Pvt. Ltd. E-7, Chand Society, Juhu Road, Juhu, Mumbai - 400 049, India

PREFACE

Water covers 70.9% of the Earth's surface and is vital for all known forms of life. Approximately 70% of the fresh water which is actively handled by humans is consumed by agriculture. Clean drinking water is essential to humans and other life forms. Access to safe drinking water has improved steadily and substantially over the last decades in almost every part of the world. There is a clear correlation between access to safe water and GDP per capita. However, some observers have estimated that by 2025 more than half of the world population will be facing water-based vulnerability. Water plays an important role in the world economy, as it functions as a solvent for a wide variety of chemical substances and facilitates industrial cooling and transportation.

This report includes water quality data for the period of **June 2010 to May 2011 (this period is known as water year)**, to know changes occurs in the selected parameters of selecting all stations of Dams & River water body which comes under Hydrology Project Division Aurangabad. The agency **M/s. Ashwamedh Engineers & Consultants Co-Op. Soc. Ltd. Aurangabad** was awarded the contract towards Operation and Maintenance of Water Quality Lab Level-II, Aurangabad for the said period. The data has been interpreted to know the affected locations.

Therefore it is a great pleasure to hand over this precise report on analysis of water samples at Water Quality Lab Level–II, Aurangabad. This booklet attempts to briefly describe an over view and general conclusion on the basis of water quality data of water samples collected from selected locations for define frequencies for the reported period.

It is expected that this booklet will provide a brief idea about Water Quality at the locations which comes under the Water Quality Lab Level -II at Aurangabad. Our efforts can always be updated through valuable suggestions.

> (V. L. Joshi) Executive Engineer Hydrology Project Division Aurangabad.

Annual Report

Water Quality Monitoring Through Water Quality Lab Level-II Aurangabad for the Year 2010-11

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CHAPTER – I EXECUTIVE SUMMARY

CHAPTER-I

EXECUTIVE SUMMERY

Annual Report

Water Quality Monitoring Through Water Quality Lab Level-II, Aurangabad for the Year 2010-2011

1.1 Preamble:

The water quality monitoring in the area of surface water is performed in order to determine the quality of water. Various parameters are analyzed in the laboratory and 6 parameters are tested at field level. All these tasks are recorded and utilized for preparing the Annual Report by performing some specific exercise. This data is considered in order to specify the quality of water at each location. This also helps to determine the pollution level or concentration in each source of water at each station.

1.2 Water Quality Monitoring – Objectives:

Observations of analysis of physical & chemical parameters are carried out as per **"Uniform Protocol for Water Quality Monitoring Order 2005"** for each location followed by Operation and Maintenance of Water Quality Laboratory Level-II, Aurangabad. As per Standard Guidelines and mandates including collection, transportation, analysis of samples, data entry in SWDES Software and preparation of the said Annual Report as per specific guidelines issued by Executive Engineer, Hydrology Project Division, Aurangabad.

1.3 Water Quality Monitoring – Scope:

Annual Report is prepared for the period from June – 2010 to May – 2011 (which is consider as Water Year period). In order to study the water quality status location wise, all stations are covered for this report which comes under this lab during the year 2010-11.

1.4 Methodology:

Analysis of Physical and Chemical parameters is done in the laboratory on the basis of Standard Analytical Methods, Instrument Operating Instructions, HIS Manuals, CPCB Guidelines and APHA 21st Ed., 2005.

Data analyzed further validated with prescribed method as per Water Quality Manuals to verify various ratios manually and is entered in SWDES Software for Water Quality Data Entry. Further the data is sent to State Data Center for further dissemination to user end.

Furthermore to get an idea of about data generated for the period it is decided and instructed to analyzed the generated data for the said period in the form of Annual Report with the help of various tools in SWDES Software to find out critical parameters and critical locations in the jurisdiction of this Lab.

1.5 Result and Observation:

In the year 2010-11, after observing all this data it is clear that most of the physical parameters are within tolerance limit except at few locations, we found that some parameter's value exceeds, mainly in stations like Aurangabad & Lonar. We have here separately exhibited the identical difference in the water of Lonar by providing detail graphs of some of the important parameters of water analysis. This will help to understand that how much water can be critical & have no use in such condition.

1.6 Conclusion:

In the Year 2010-11 it can be concluded that all the parameters of stations are in tolerance limit except some stations like Aurangabad & Lonar, which are observed polluted. When fresh water is artificially supplemented with nutrients, it results in an abnormal increase in the growth of water plants. This is known as eutrophication. The discharge of waste from industries, agriculture and urban communities into water bodies generally stretches the biological capacities of aquatic systems. Chemical run-off from fields also adds nutrients to water. Excess nutrients

cause the water body to become choked with organic substances and organisms. When organic matter exceeds the capacity of the microorganisms in water that break down and recycle the organic matter, it encourages rapid growth or blooms of algae. When they die, the remains of the algae add to the organic wastes already in the water, eventually the water becomes deficient in oxygen. Anaerobic organisms (those that do not require oxygen to live) then attack the organic wastes, releasing gases such as methane and hydrogen sulphide, which are harmful to the oxygen-requiring (aerobic) forms of life. Hence it can be concluded that water from all these sources have pollution and so it requires treatment before its usage.

In the point of consideration for above locations, water is suitable for irrigation purpose followed by traditional irrigation method.

1.7 Recommendations/Remedial Measures:

- Domestic effluents may be treated and disinfected before discharging.
- Effluents from the non-point sources may be identified. These are required to be collected and treated.
- Use of water of such polluted locations may be useful for salt tolerance crop and is recommended based on special study.
- Use of direct source of water is to be avoided.
- Bathing at such location should be restricted.

1.8 Suggestions:

- Create mass awareness in general public regarding surface and ground water quality aspects.
- Water Quality Annual Report shall be publicly published every year and made available to everyone on demand.
- Sampling points of sampling location need to be increased to better analysis of data at different stages of that location.

1.9 Water Quality Trend in Aurangabad Division & Amaravati Division

Water Quality Trend in Aurangabad & Amaravati division is based on the data available from June 2010 to May 2011 in Water Quality Lab Level-II, Aurangabad. This comparison of change is related to Irrigation parameters on their concentration.

CHAPTER -II INTRODUCTION

CHAPTER-II

INTRODUCTION

2.1 General:

The water quality monitoring in the area of surface water is performed in order to determine the quality of water. Various parameters are analyzed in the laboratory and 6 parameters are tested at field level. All these tasks are recorded and utilized for preparing the Annual Report by performing some specific exercise. These data are considered in order to specify the quality of water at each location. This also helps to determine the pollution level or concentration in each source of water at each station.

2.2 Water Quality Monitoring – Objectives:

Observations of analysis of physical & chemical parameters are carried out as per **"Uniform Protocol for Water Quality Monitoring Order 2005"** for each location followed by Operation and Maintenance of Water Quality Laboratory Level-II, Aurangabad. As per Standard Guidelines and mandates including collection, transportation, analysis of samples, data entry in SWDES Software and preparation of the said Annual Report as per specific guidelines issued by Executive Engineer, Hydrology Project Division, Aurangabad.

2.3 Water Quality Monitoring – Scope:

The Annual Report is prepared for the year 2010-11. The Table below shows the number of sample analyzed during the reported period. In order to study water quality status station wise, all locations covered under this lab during the year 2010-11 are considered.

Seasonal averages of all analyzed parameters are calculated for study of seasonal water quality trend at each location.

2.4 Other Activities:

Apart from working for regular Water Quality Monitoring for Water Quality lab level II at Aurangabad, the infrastructure facility is made available to the users from various Government, Non Government, Private sector as well as individuals.

The facility is availed by many users with testing of sample towards drinking purpose, ice factory, construction purpose, swimming tanks, irrigation purpose & study purpose, also in plant training facility for Students of Educational Institutes.

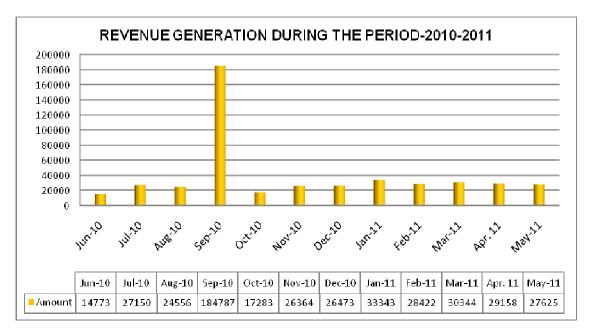
During the year 2010-11 many clients approached to the laboratory. The valuable clients availed the facility of the laboratory are as below;

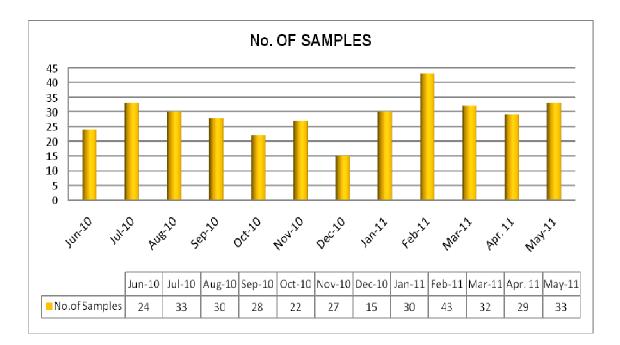
REVENUE GENERATED DURING THE REPORTING PERIOD (June 2010 – May 2011)

Sr. No.	Water Year	No. of Samples Analyzed	Amount Received
1.	2010-2011	346	4,70,278

Month wise details are as under

Month & Year	Total Amount	No of Sample
Jun-10	14,773	24
Jul-10	27,150	33
Aug-10	24,556	30
Sep-10	1,84,787	28
Oct-10	17,283	22
Nov-10	26,364	27
Dec-10	26,473	15
Jan-11	33,343	30
Feb-11	28,422	43
Mar-11	30,344	32
Apr. 11	29,158	29
May-11	27,625	33
Total:	4,70,278	346





2.5 Extended Scope of Laboratory:

Water Quality Lab Level – II is certified by **ISO 9001-2008** in the year 2009-10 and we are working as per the standards of ISO to maintain the quality of work to streamline the process of work to achieve the set objectives.

CPCB AQC Exercise:

In the month of Feb. 2011 CPCB conducted the AQC Exercise, in which this lab score 81.72%.

Training and Visits:

Infrastructure facility is made available to many college students for in plant training and their research purposes. Many visitors from various institutes like Dr. B. A. M. University, Aurangabad, JNEC College Aurangabad, students of Savitribai Phule B.Ed. College & students of various schools visited the laboratory.

The infrastructure facility is also made available to all institutional organization for visit and study purposes and the generated data of Water Quality is also made available to the users who are a member of Hydrology Project.

WATER QUALITY LABORATORY LEVEL-II @ AURANGABAD

SALIENT FEATURES OF LABORATORY

1.	Latitude	: 19° 51″30″
2.	Longitude	: 75° 21″18″
3.	River Basin	: Godavari
4.	Year of Establishment	: 2001
5.	Actual Working Started	: October-2001
6.	Sampling Locations As	
	Per W.Q. Network covered in this	lab. : 36
7.	Monthly sample collection	: 43 samples
8.	Samples Frequencies	
	a) Baseline	:
	b) Trend/Flux	: Monthly
	c) Reservoir	: Fortnightly
9.	Level of Lab	: Level II
10.	W.Q. Parameters for Level II Lab	: 39 Nos.
11.	Observation Frequency	: Daily
12.		1) Mr. V. D. Nemade (Ex. Engineer) 2) Mr. V. P. Kulkarni (S.D.O) 3) Mr. A. D. Mahajan (A.E. II) 4) Mrs. S. S. Quadri (Silt Analysist)
13.		Ashwamedh Engnrs. & Cons. C. S. L. Aurangabad Branch.
14.	Staff Position :	 Ms. V. P. Pawar (Chief Chemist) Ms. S. P. Deshpande (Microbiologist) Ms. Pushpa Patil (Chemist) Mr. S. R. Wankhede (Field Chemist) Mr. S. Gangawane (Lab Assistant)

SCOPE OF WORK: OPERATION AND MAINTENANCE OF WATER QUALITY LABORATORY LEVEL-II AURANGABAD.

Outdoor work: Surface water sampling from selected water quality network sampling points as per schedule of sampling, to be collected as per norms given by World Bank including field test to be carried out on site as per specific Performa of sampling to be collected from laboratory and sample should be reached in the lab, within 24 hrs with field determination test as specified by lab in charge etc.

The Surface Water sampling includes:

- Field determination as per standard guideline.
- Field parameters to be tested on site & entry to be taken on ID form.
- Sample to be transported to laboratory within prescribed time limit. Work is carried out as per flow chart.

Transportation: Transportation of collected samples from selected sampling points as per list including transportation sampling material as per check list from lab and transportation samples from sampling points back to laboratory as per schedule within 24 hrs.

Indoor work:- Operation & Maintenance of Water Quality Laboratory Level-II @ Aurangabad including analysis of water samples as per test procedure & taking entry of collected samples, operating instruments as per manual & keeping data record, primary & secondary validation, participating in AQC Exercise, analysis of samples under AQC, operation within Laboratory AQC exercise including analysis of samples within 30/23 parameters for Dam Samples and 28/21 parameters for trend samples as directed by Lab Incharge each from every station per month within stipulated period by using scientific staff including instrument operations data record maintenance & laboratory Management etc complete.

Steps in Indoor work:

- Day to Day Operation and Maintenance of Water Quality Laboratory Level II.
- The work includes analysis of water samples as per the test procedures.
- Operating the instruments as per specified instruction manual.
- Entry of data in SWDES Software.
- Conducting Analytical Quality Control Exercise (AQC) round.
- Within Laboratory AQC ii) Intra Laboratory AQC The Laboratory staff employed;
- Chief Chemist: 1 No.
- Sr. Research Officers: 1 No.
- Research Assistant: 2 No.
- Lab. Assistant: 1 No.
- The Indoor work also includes keeping data record.
- Log book of Lab equipment
- Preparation of monthly sampling Schedule.
- Keeping sampling record, instruments operation, Laboratory Management, demonstration
- Training to Departmental staff as and when required.
- Information to visitors & Customer

Water Quality Lab Level-II @ Aurangabad. Statement Showing Number of Samples proposed to be collected for the Period of Work- June 2010 to May-2011

The Water Quality Laboratory Level-II at Aurangabad is taken in Operation and Maintenance and data for the period of June 2010 to May 2011 is taken for report preparation. Total no of 36 sampling location covered, out of which 28 locations are from old river stations of Aurangabad division & Amaravati division, 7 Dam Stations & 1 is creator station. Actual sample from Pategaon Trend Station is collected from Jayakwadi Project.

Table showing No. of Locations Covered under the Jurisdiction ofWater Quality Lab level – II, Aurangabad.

Sr.	Name & Type of Sampling Stations			Monthly	Remark		
No.	Туре	Station	Name of River	Frequency of sampling			
	Location as per W.Q. Network						
Aura	ngabad [Division: River Loc	ation				
1	Trend	Killari	Terna	Monthly			
2	Trend	Takali Udgir	Manjara	Monthly			
3	Trend	Newasa	Godavari	Monthly			
4	Trend	Wadvali	Godavari	Monthly			
5	Trend	Tokabridge	Pravra	Monthly			
6	Trend	Shirur	Sindaphana	Monthly			
7	Trend	Patoda	Manjara	Monthly			
8	Trend	Raheri	Purna	Monthly			
9	Trend	Kesrali	Manar	Monthly			
10	Trend	Sundagi	Lendi	Monthly			
11	Trend	Nandednaga	Godavari	Monthly			
12	Trend	Yelli	Godavari	Monthly			
13	Trend	Purnabridge	Purna	Monthly			
14	Trend	Takalidhangar	Godavari	Monthly			
15	Trend	Hirapur	Sindaphana	Monthly			
16	Trend	Shahgad	Godavari	Monthly			
17	Trend	Aurangabad	Kham	Monthly			
18	Trend	Shendurwada	Kham	Monthly			
19	Trend	Parli Vaijanath	Wan	Monthly			
20	Trend	Gangakhed	Godavari	Monthly			
21	Trend	Latur	Manjara	Monthly			
22	Trend	Auradshahjahani	Lower Terna	Monthly			
23	Trend	Rahata	Purna	Monthly			
24	Trend	Ajanta-Andhari	Andhari	Monthly			

Ama	Amarvati Division: River Location					
25	Trend	Padalse	Тарі	Monthly		
26	Trend	Bhusaval	Тарі	Monthly		
27	Trend	Savkheda	Girna	Monthly		
28	Trend	Hingona	Mor	Monthly		
Aura	angabad D	ivision: Dam Loc	ation			
29	Dam	Lower Terna	Terna	Fortnightly		
30	Dam	Manjara	Manjara	Fortnightly		
31	Dam	Majalgaon	Sindaphana	Fortnightly		
32	Dam	Yeldari	Purna	Fortnightly		
33	Dam	Vishnupuri	Godavari	Fortnightly		
34	Dam	U.P.P.	Penganga	Fortnightly		
35	(Trend / Dam)	Pategaon	Godavari	Fortnightly	Sample is collected from Jayakwadi project	
36	Dam	Lonar Creator	Natural Creator	Fortnightly		

Statement Showing Water Quality Parameter Tested at W.Q. Lab Level-II Aurangabad.

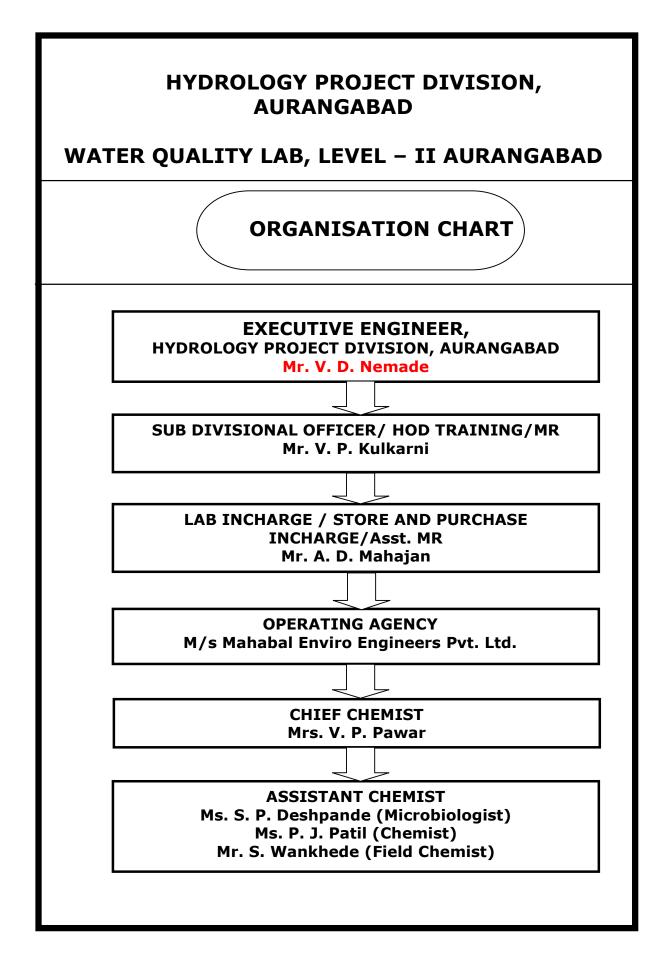
	First Round	Second Round	
Sr. No. Parameters		Parameters	
1	Colour	Colour	
2	Odour	Odour	
3	Temperature	Temperature	
4	рН	рН	
5	Electrical Conductivity	Electrical Conductivity	
6	DO	DO	
7	Turbidity	Turbidity	
8	Total Solids	Total Solids	
9	Dissolved Solids	Dissolved Solids	
10	Suspended Solids	Suspended Solids	
11	Ammonia NH ₃	Ammonia NH ₃	
12	Nitrite NO ₂	Nitrite NO ₂	
13	Nitrate NO ₃	Nitrate NO ₃	
14	T Phosphorous	T Phosphorous	
15	B.O.D	B.O.D	
16	C.O.D	C.O.D	
17	Potassium K	Chloride Cl	
18	Sodium Na	Total coliforms	
19	Magnesium Mg	Fecal Coliforms	
20	Carbonate CO ₃	Alkalinity	
21	Bicarbonate HCO ₃	Sodium Na	
22	Calcium		
23	Chloride Cl		
24	Fluoride F		
25	Boron B		
26	Total coliforms		
27	Fecal Coliforms		
28	Alkalinity		

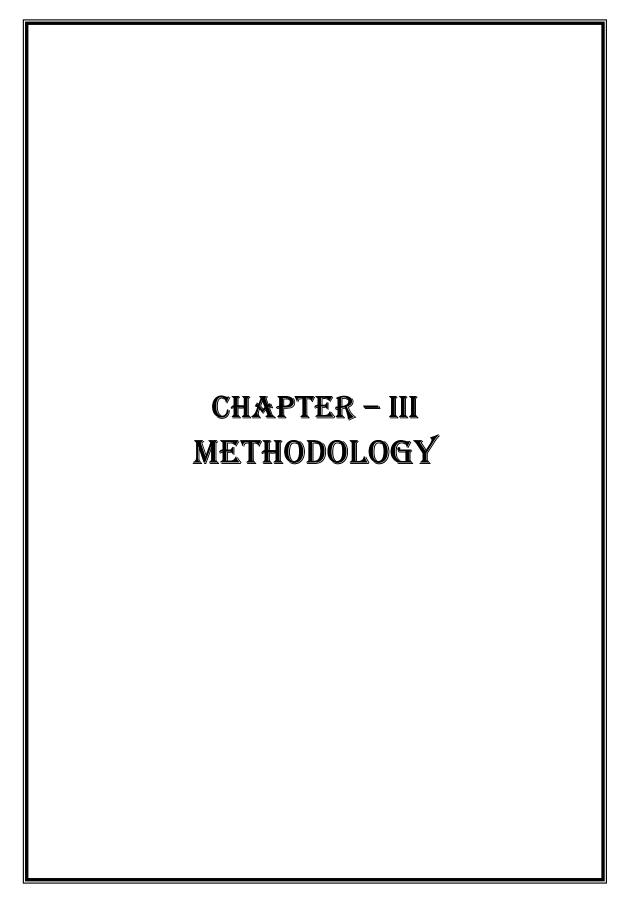
Trend & Flux

Statement Showing Water Quality Parameter Tested at W.Q. Lab Level-II Aurangabad.

Dam Sample (Reservoir)

	First Round	Second Round
Sr. No.	Parameters	Parameters
1	Colour	Colour
2	Odour	Odour
3	Temperature	Temperature
4	рН	рН
5	Electrical Conductivity	Electrical Conductivity
6	DO	DO
7	Turbidity	Turbidity
8	Total Solids	Total Solids
9	Dissolved Solids	Dissolved Solids
10	Suspended Solids	Suspended Solids
11	Ammonia NH ₃	Ammonia NH ₃
12	Nitrite NO ₂	Nitrite NO ₂
13	Nitrate NO ₃	Nitrate NO ₃
14	T Phosphorous	T Phosphorous
15	B.O.D	B.O.D
16	C.O.D	C.O.D
17	Pottassium K	Chloride Cl
18	Sodium Na	Total coliforms
19	Magnesium Mg	Fecal Coliforms
20	Carbonate CO ₃	Alkalinity
21	Bicarbonate HCO ₃	Sodium Na
22	Calcium	Kjeldahl Nitrogen
23	Chloride Cl	Chlorophyll-a
24	Fluroide F	
25	Boron B	
26	Total coliforms	
27	Fecal Coliforms	
28	Alkalinity	
29	Kjeldahl Nitrogen	
30	Chlorophyll-a	





CHAPTER-III

METHODOLOGY

3.0 General:

This Water Quality laboratory covers Surface Water component like Rivers Godavari, Purna, Manjara, Terna and Reservoir like Dams & Lonar Creator from different district locations.

3.1 Rivers:

A river is a natural watercourse, usually freshwater, flowing toward an ocean, a lake, a sea or another river. In a few cases, a river simply flows into the ground or dries up completely before reaching another body of water. Small rivers may also be called by several other names, including stream, creek, brook, rivulet and rill.

Fortunately almost the entire country is criss-crossed by rivers. Geographical area of the state is divided in different river basins viz. Godavari, Purna, Manjara, Terna etc.

3.2 Methodology

Analysis of Physical and Chemical parameters is done in the laboratory on the basis of Standard Analytical Methods, Instrument Operating Instructions, HIS Manuals, CPCB Guidelines and APHA, 21st Ed., 2005. Data analyzed further validated with prescribed method as per Water Quality Manuals to verify various Ratios manually and is entered in SWDES Software for Water Quality Data Entry. Further the data is sent to State Data Center for further dissemination to user end.

Furthermore to get an idea of about data generated for the period it is decided and instructed to analyzed the generated data for the said period in the form of Annual report with the help of various tools in SWDES Software to find out critical parameters and critical locations in the jurisdiction of this Lab.

Methodology For the analysis of Water Quality samples the following parameters were analyzed during the Period 2010-11

Table showing List of parameters and the methodology used for analysis.

Sr. No	Parameters	Methodology
1.	Colour	APHA, 21 st Ed., 2005, 2120-B, 2-2
2.	Odour	IS 3025 (Part 5): 1983, Reaffirmed 2006
3.	Temperature	APHA, 21 st Ed., 2005, 2550-B, 2-61
4.	pH	APHA, 21 st Ed., 2005, 4500-H ⁺ - B, 4-90
5.	Electric Conductivity	APHA, 21 st Ed., 2005, 2510- B, 2-47
6.	Dissolved Oxygen	APHA, 21 st Ed., 2005, 4500-O, 4138
7.	Turbidity	APHA, 21 st Ed., 2005, 2130-B, 2-9
8.	Total Solids	APHA, 21 st Ed,2005, 2540 B, -266
9.	Dissolved Solids	APHA, 21 st Ed,2005, 2540 C, 251
10.	Suspended Solids	APHA, 21 st Ed,2005, 2454 D-258
11.	NH ₃ -N	APHA, 21 st Ed., 2005, 4500-NH ₃ F, 4-110
12.	NO ₂ ⁻	APHA, 21 st Ed., 2005, 4500-NO ₂ -B, 4-118
13.	NO ₃ ⁻	APHA,21 st Ed., 2005, 4500-NO ₃ , B -4 -120
14.	Total Phosphorous	APHA, 21 st Ed., 2005, 4500 P, E, 4-153
15.	Biochemical Oxygen Demand	APHA, 21 st Ed., 2005, 5210-52
16.	Chemical Oxygen Demand	APHA, 21 st Ed., 2005, 5220-B, 5-15
17.	Potassium K ⁺	APHA, 21 st Ed., 20053500K,388
18.	Sodium Na ⁺	APHA, 21 st Ed., 2005, 3500NA,398
19.	Calcium Ca ⁺⁺	APHA, 21 st Ed., 2005, 3500-B, 3-65
20.	Magnesium Mg ⁺⁺	APHA, 21 st Ed., 2005, 3500-Mg, B, 3-84
21.	Iron (as Fe)	APHA, 21 st Ed., 2005, 3111-B, 3-17
22.	Carbonate CO ₃	APHA, 21 st Ed., 2005, 2320-B, 2-27, 5 -1 & 4500-CO ₂ -D, 4-34
23.	Bi-Carbonate H CO ₃	APHA, 21 st Ed., 2005, 2320-B, 2-27, 5 -3 & 4500-CO ₂ -D, 4-34
24.	Chloride Cl	APHA, 21 st Ed., 2005, 4500-Cl, B, 4-70
25.	Fluoride F	APHA, 21 st Ed., 2005, 4500-F ⁻ , D, 4-85
26.	Boron B	APHA, 21 st Ed., 2005, 4500-B-C, 4-23
27.	Total Coliforms	APHA, 21 st Ed., 2005, 9221-B, 9-49
28.	Faecal Coliforms	APHA, 21 st Ed., 2005, 9221-E, 9-56
29.	Alkalinity	IS 3025 (Part 23): 1986, Reaffirmed 2003, Amds.1

3.4 Flow Chart:

The work of analysis of sample is being monitored on the basis of flow chart generated in the lab as per standard guidelines and analysis of sample is performs as per guidelines of world bank with HIS manuals and APHA ,21 st Ed, 2005 as a standard procedures for analysis of samples. As well refers BIS standards IS:10500 and other relevant BIS standards for analysis of various samples received from users for various purpose like drinking, irrigation, bathing (swimming tank), construction, study & various R & D activities.

FLOW CHART OF ANALYSIS OF HP WATER SAMPLE

Sample Collection from Sampling Source with the help of Depth Sampler

Treatment: D.O. Fixing, Preservation of MPN Sample, Colour, Odour Temp, pH, EC are tested on field, and fill up in ID form.

At Laboratory: Inward the Sample, Giving the Sr. No. to the sample noted into sample entry register

ID form entry taken into SWEDS Software

Tests are carried out in lab as per Protocols. These tests are: Microbiological test, Chlorophyll-a, Temp, pH, D.O., B.O.D, Ammonia, Nitrate, Nitrite, TDS, TSS, C.O.D., Turbidity, Alkalinity, Carbonate & Bicarbonate, Chloride, Fluoride, Boron, Iron, Sodium, Potassium, Total Hardness, Phosphorous, and Calcium etc.

Observations & Calculations of all Analyzed Parameters are entered in the Data Sheet

The results of parameters are checked & validated

After Validation Check, all the data is entered in to Data Record and Validation Register

This data is finally entered in to SWEDS Software

This data is submitted to Hydro Metrological Division Nashik for further action

FLOW CHART OF ANALYSIS OF NHP WATER SAMPLE

Sample Collection from Party/Person

Sample forms fill up and issuing receipt of cash received.

Inward the Sample, Giving the Sr. No. to the sample noted into sample entry register

Sample Analysis as per the customer's requirement

Observations & calculations of all Analyzed Parameters

The results of parameters are checked & prepared

Issue of Final Result to Customer

Annual Report On Water Quality Monitoring through Water Quality Lab level II. Aurangabad for the year 2010-11

TABLE SHOWING SAMPLES ANALYSED DURING THE REPORTING PERIOD

Year	Trend Sample (First Round)	Trend Sample (Balance Round)	Dam Sample (First Round)	Dam Sample (Balance Round)	Total
2010-11	28	96	8	183	315

CHAPTER – IV RESULT & OBSERVATIONS

CHAPTER - IV

RESULTS AND OBSERVATIONS

4.0 Results and Conclusions:

The water quality monitoring in the area of surface water is performed in order to determine the quality of water. Various parameters are analyzed in the laboratory and 6 parameters are tested at field level. All these tasks are recorded are utilized for preparing the Annual Report by performing some specific exercise. These data are considered in order to specify the quality of water at each location. This also helps to determine the pollution level or concentration in each source of water at each station.

4.1 Water Quality status- Stations wise Exercise:

In order to study water quality status station wise, all locations are considered which comes under the jurisdiction of this lab during the year 2010-11.

4.2 Objectives:

Observations of respective parameters in view with use of water i.e. for drinking purpose or irrigation purpose, analyzed for each location individually & interpretation of data has done to identify the trend at that location. Also critical parameters are identified at every location.

4.3 Irrigation Water Quality Criteria:

Soil scientists use the following categories to describe irrigation water effects on crop production and soil quality:

- Salinity hazard total soluble salt content
- Sodium hazard relative proportion of sodium (Na⁺) to calcium (Ca²⁺) and magnesium (Mg²⁺) ions

- pH
- Alkalinity carbonate and bicarbonate

Specific ions: chloride (Cl), sulfate (SO4²⁻), boron (B), and nitratenitrogen (NO3^{-N}).

Other potential irrigation water contaminants that may affect suitability for agricultural use include heavy metals and microbial contaminants

Sr. No.	Parameter	limit	Unit
1.	рН	6.5-8.5	-
2.	Electrical Conductivity	2250	Micromhos/cm
3.	Total Dissolved Solids	2100	mg/Lit
4.	Chloride	600	mg/Lit
5. Sulphate		1000	mg/Lit
6. Boron		2	mg/Lit
7.	% Sodium	60	%
8.	SAR (Sodium Absorbance Ratio)	26	-

ICAR Standard for Irrigation Water

4.4 Effects of water pollution:

The effects of water pollution are not only devastating to people but also to animals, fish and birds. Polluted water is unsuitable for drinking, recreation, agriculture and industry. It diminishes the aesthetic quality of lakes and rivers. More seriously, contaminated water destroys aquatic life and reduces its reproductive ability. Eventually, it is a hazard to human health. Nobody can escape from the effects of water pollution.

4.5 Critical parameters Identified:

After observing all this data it is clear that most of the physical parameters are within tolerance limit, Physical parameters like TDS and Chloride are within tolerance limits of Dam Samples & some of the River samples but in case of few stations it is observed that the value of TDS is increased. Most of the stations from Aurangabad Division & Amaravati Division, tested parameters are within tolerance limit as per Irrigation ICAR Standards. Most of the chemical parameters are also within tolerance limit.

OBSERVATION ON THE BASIS OF CLASSIFICATION OF LOCATION			
Sr. No.	River	Year	Observations
1	Aurangabad Division – Trend (River) Station (24)	2010-11	Water is good for irrigation without any further treatment & it belongs to C2 & S1, but at station wadwali & Nanded nagapur it is high, belongs to C3 & S1, so it can be use after treatment. Except Aurangabad
2	Amaravati Division - Trend (River) Station (4)	2010-11	Water is good for irrigation without any further treatment & it belongs to C2 & S1.
3	Dams Stations (7 +1)	2010-11	Water is good for irrigation without any further treatment & it belongs to C2 & S1. Except Lonar.

4.6 Classification of locations on the basis of Wilcox diagram for the year 2010-11

Wilcox diagram

Details of Sample Collection Stations W.R. Stations under W.R. Division, Amaravati. For the year 2010-11

From the above diagram it can be observed that, all station's water belongs to $C_2 \& S_1$ class. Hence, it can be concluded that the water can be used for irrigation purpose without any further treatment.

Details of Sample Collection Stations W.R. Stations under W.R. Division, Aurangabad. For the year 2010-11

From the above diagram it can be observed that, above station's water belongs to C_2 & S_1 class except station at Nanded nagapur & Wadwali belongs to C_3 & S_1 Class, hence it can be concluded that the water can be used for irrigation purpose without any further treatment. Except Wadwali & Nandednagapur station it requires treatment before usage.

Details of Sample Collection Stations W.R. Stations under W.R. Division, Aurangabad. Dam Location for the year 2010-11

From the above diagram it can be observed that, all station's water belongs to $C_2 \& S_1$ class. Hence, it can be concluded that the water can be used for irrigation purpose without any further treatment.

Details of Sample Collection Stations W.R. Stations under W.R. Division, Aurangabad. New Station for the year 2010-11

From the above diagram it can be observed that, above station's water belongs to C2 & S₁, C3 & S1class except Aurangabad (C3 & S1class), hence it can be concluded that the water can be used for irrigation purpose without any further treatment but in Aurangabad it requires treatment before usage.

Graphical Representation of Boron (As per ICAR Standards Parameter) for the Year 2010-2011 WR Division Aurangabad



Boron is another element that is essential in low amounts, but toxic at higher concentrations. In fact, toxicity can occur on sensitive crops at concentrations less than 2.0 ppm.

From above station's graph it is observed that Boron is within tolerance limit at all stations.

Graphical Representation of Chloride (As per ICAR Standards Parameter) for the Year 2010-2011 WR Division Aurangabad



Chloride is a common ion in irrigation waters. Although chloride is essential to plants in very low amounts, it can cause toxicity to sensitive crops at high concentrations, like sodium, high chloride concentrations cause more problems.

From above graph it is observed that at these stations maximum chloride is found at Aurangabad, Wadvali & Shendurwada stations, as per ICAR Stds., it is within limit.

Graphical Representation of Electrical Conductivity (As per ICAR Standards Parameter) for the Year 2010-2011 WR Division Aurangabad

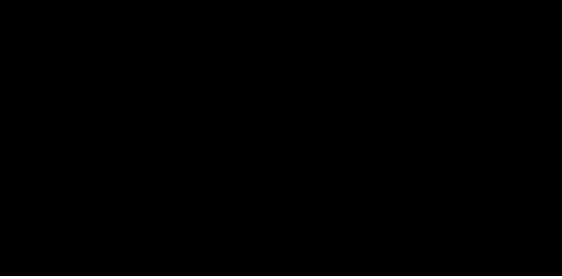




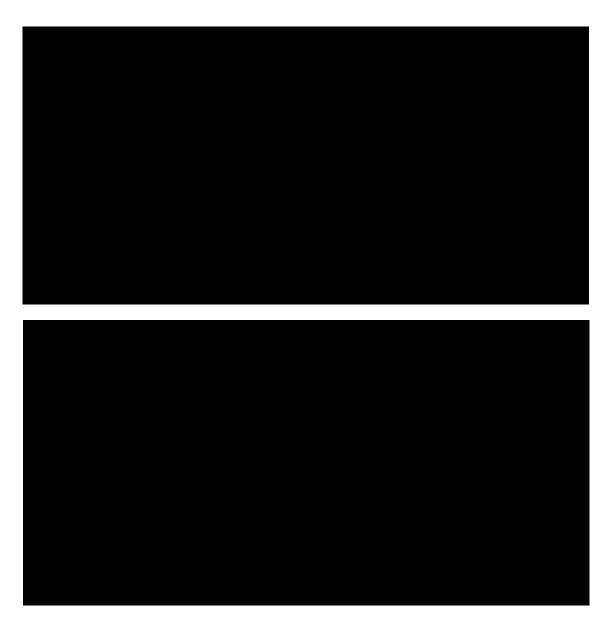
Electrical conductivity is an indicator of dissolved metals. Some common metals that may be found in surface water include iron, aluminum, calcium, magnesium and others. High conductivity levels may be due to several different factors.

From above graph it is observed that maximum Electrical conductivity found at Aurangabad station. As per ICAR Stds water at all stations is within tolerance limit. Graphical Representation of Na% (As per ICAR Standards Parameter) for the Year 2010-2011 WR Division Aurangabad





From above Graph it is observed that the Na% value has fluctuation at every station. However many factors including soil texture, organic matter, crop type, climate, irrigation system and management impacts on how sodium in irrigation water affects soils. Though as per ICAR standards it is within limit, except Wadwali. Graphical Representation of pH (As per ICAR Standards Parameter) for the Year 2010-2011 WR Division Aurangabad



The acidity or basicity of irrigation water is expressed as pH (< 7.0 acidic; > 7.0 basic). High pH's above 8.5 are often caused by high bicarbonate (HCO_3^{-}) and carbonate (CO_3^{-2}) concentrations, known as alkalinity. From above graph it is observed that as per ICAR standards water of all stations is not within tolerance limit.

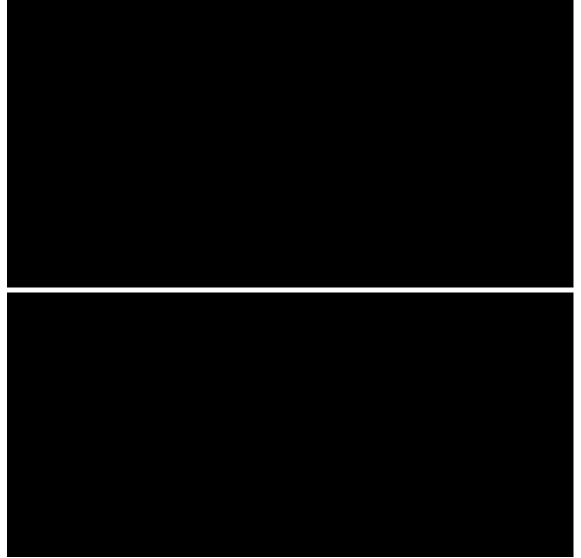
Graphical Representation of SAR (As per ICAR standards Parameter) for the Year 2010-2011 WR Division Aurangabad



SAR fluctuate while EC is an assessment of all soluble salts in a sample, sodium hazard is defined separately because of sodium's specific detrimental effects on soil physical properties. The sodium hazard is typically expressed as the sodium adsorption ratio (SAR). This index quantifies the proportion of sodium (Na⁺) to calcium (Ca⁺⁺) and magnesium (Mg⁺⁺) ions in a sample. Calcium will flocculate (hold together), while sodium disperses (pushes apart) soil particles. This dispersed soil will readily crust and have water infiltration and permeability problems.

As per above graph SAR is within tolerance limit.

Graphical Representation of Total Dissolved Solids (As per ICAR Standards Parameter) for the Year 2010-2011 WR Division Aurangabad



TDS is a general indicator of overall water quality. It is a measure of inorganic and organic materials dissolved in water. High levels of TDS in surface water may be due to several factors, including sedimentation, mining or storm water runoff. Increased TDS may impart a bad odor or taste to drinking water.

From above graph it is observed that the TDS value at all stations is not within tolerance limit, at some locations like Aurangabad, Shendurwada, Toka & Shahagad, it is on much higher side. Graphical Representation of Boron (As per IS10500 Standards) for the Year 2010-2011 WR Division Aurangabad Dam Locations



Boron is another element that is essential in low amounts, but toxic at higher concentrations. In fact, toxicity can occur on sensitive crops at concentrations less than 2.0 ppm.

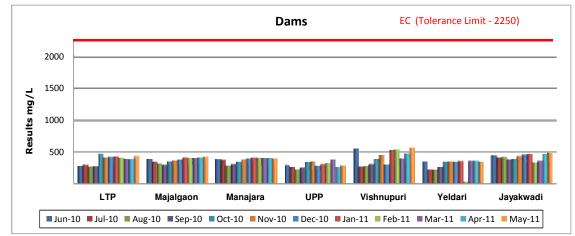
From above station's graph it is observed that Boron is within tolerance limit.

Graphical Representation of Na% (As per IS10500 Standards) for the Year 2010-2011 WR Division Aurangabad Dam Locations



From above Graph it is observed that the Na% value has fluctuation at every station. However many factors including soil texture, organic matter, crop type, climate, irrigation system and management impacts on how sodium in irrigation water affects soils. Though as per IS-10500 standards it is within limit.

Graphical Representation of Electrical Conductivity (As per IS10500 Standards) for the Year 2010-2011 WR Division Aurangabad Dam Locations



Electrical conductivity is an indicator of dissolved metals. Some common metals that may be found in surface water include iron, aluminum, calcium, magnesium and others. High conductivity levels may be due to several different factors.

From above graph it is observed that as per IS-10500 Stds water at all station's Electrical Conductivity is within tolerance limit.

Graphical Representation of pH (As per IS10500 Standards) for the Year 2010-2011 WR Division Aurangabad Dam Locations



The acidity or basicity of irrigation water is expressed as pH (< 7.0 acidic; > 7.0 basic). High pH's above 8.5 are often caused by high bicarbonate (HCO_3^-) and carbonate (CO_3^2 -) concentrations, known as alkalinity. From above graph it is observed that as per IS-10500 standards water of all station's pH is within tolerance limit, except LTP & Jayakwadi. Graphical Representation of SAR (As per ICAR Standards) for the Year 2010-2011 WR Division Aurangabad Dam Locations



SAR fluctuate while EC is an assessment of all soluble salts in a sample, sodium hazard is defined separately because of sodium's specific detrimental effects on soil physical properties. The sodium hazard is typically expressed as the sodium adsorption ratio (SAR). This index quantifies the proportion of sodium (Na⁺) to calcium (Ca⁺⁺) and magnesium (Mg⁺⁺) ions in a sample. Calcium will flocculate (hold together), while sodium disperses (pushes apart) soil particles. This dispersed soil will readily crust and have water infiltration and permeability problems.

As per above graph SAR is within tolerance limit.

Graphical Representation of TDS (As per IS10500 Standards) for the Year 2010-2011 WR Division Aurangabad Dam Locations



As per drinking Standards of TDS limit (500 mg/L), from above graph it is observed that in every month TDS of water is within tolerance limit. TDS is a general indicator of overall water quality. It is a measure of inorganic and organic materials dissolved in water. High levels of TDS in surface water may be due to several factors. Graphical Representation of FCol-MPN (As per IS10500 Standards) for the Year 2010-2011 WR Division Aurangabad Dam Locations



When a sample is positive for T coliform, it means there is fecal contamination in the water. This could be human or animal, but its implications are more serious than total coli form.

From above graph it is found that at every station there is Fecal Contamination in the water in various seasons. It needs chlorination treatment before use. Graphical Representation of TCol-MPN (As per IS10500 Standards) for the Year 2010-2011 WR Division Aurangabad Dam Locations



At the time of testing for bacteria two results are given as total Coliform. The first, total coliform is a test for coliform bacteria which are distributed widely in the environment. They are on animals, plants and in the solid, mainly in large numbers on the feces of warm-blooded animals. A positive reading for total coliform bacteria means the water has been affected by the environment and disease-causing organism may or may not be present.

From above graph it is observed that at every station there is Bacterial count in every month. Water should be treated before use.

Graphical Representation of Boron (As per ICAR Standards) for the Year 2010-2011 Amaravati Division



As per above station's graph it is observed that Boron is within tolerance limit. Boron is another element that is essential in low amounts, but toxic at higher concentrations. In fact toxicity can occur on sensitive crops at concentrations less than 2.0 ppm. Soils and irrigation waters contain enough Boron, that additional Boron fertilizer is not required in most situations. Because Boron toxicity can occur at such low concentrations, an irrigation water analysis is advised for ground water before applying additional Boron to crops. Graphical Representation of Chloride (As per ICAR Standards) for the Year 2010-2011 Amaravati Division



The secondary drinking water standard for chloride is 250 mg/L, because some people can detect a salty taste when chloride exceeds 250 mg/L. Chloride itself has no health effect. However, the amount of chloride can be directly tied to the amount of sodium in the water. From graph it is observed that at all stations it is within tolerance limit.

Graphical Representation of Electrical Conductivity (As per ICAR Standards) for the Year 2010-2011 Amaravati Division



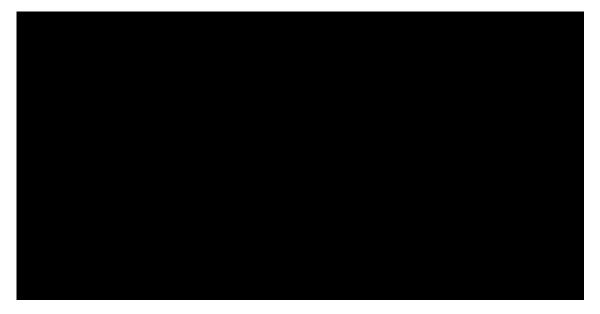
Electrical conductivity is an indicator of dissolved metals. Some common metals that may be found in surface water include iron, aluminum, calcium, magnesium and others. High conductivity levels may be due to several different factors.

From above graph it is observed that as per ICAR Stds water at all stations is within tolerance limit.

Graphical Representation of Na% (As per ICAR Standards) for the Year 2010-2011 Amaravati Division



From above Graph it is observed that the Na% value has fluctuation at every station. However many factors including soil texture, organic matter, crop type, climate, irrigation system and management impacts on how sodium in irrigation water affects soils. But as per ICAR standards it is within limit Graphical Representation of pH (As per ICAR Standards) for the Year 2010-2011 Amaravati Division



pH means the acidity or basicity of water. From graph we observed that pH value is in range of 8.1 to 8.3. The normal pH range for irrigation water is from 6.5 to 8.5. Abnormally low pH's are not common in Colorado, but may cause accelerated irrigation system corrosion where they occur. High pH's above 8.5 are often caused by high bicarbonate (HCO3⁻) and carbonate (CO3²⁻) concentrations, known as alkalinity.

Graphical Representation of SAR (As per ICAR Standards) for the Year 2010-2011 Amaravati Division



SAR fluctuate while EC is an assessment of all soluble salts in a sample, sodium hazard is defined separately because of sodium's specific detrimental effects on soil physical properties. The sodium hazard is typically expressed as the sodium adsorption ratio (SAR). This index quantifies the proportion of sodium (Na⁺) to calcium (Ca⁺⁺) and magnesium (Mg⁺⁺) ions in a sample. Calcium will flocculate (hold together), while sodium disperses (pushes apart) soil particles. This dispersed soil will readily crust and have water infiltration and permeability problems.

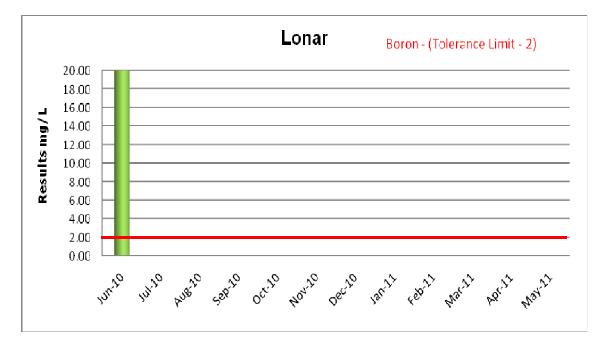
As per above graph SAR is within tolerance limit.

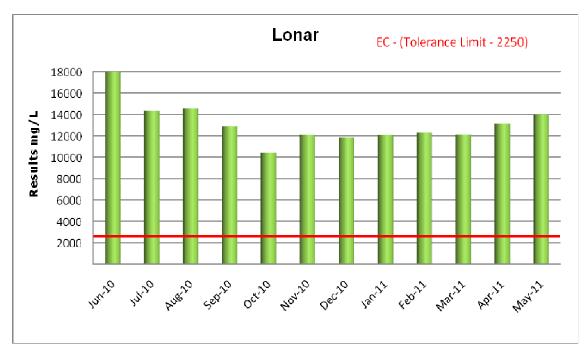
Graphical Representation of Total Dissolved Solids (As per ICAR Standards) for the Year 2010-2011 Amaravati Division



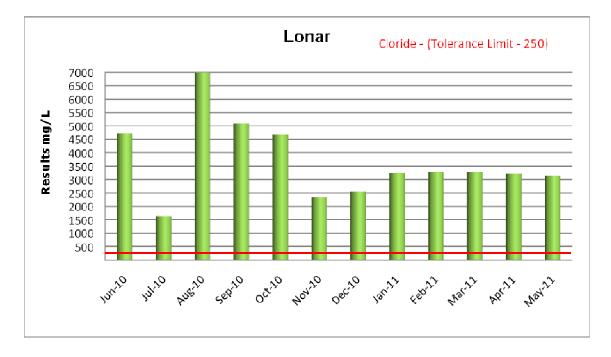
As per drinking Standards of TDS limit (500 mg/L), from above graph it is observed that in every month TDS of water is within limit. TDS is a general indicator of overall water quality. It is a measure of inorganic and organic materials dissolved in water. High levels of TDS in surface water may be due to several factors.

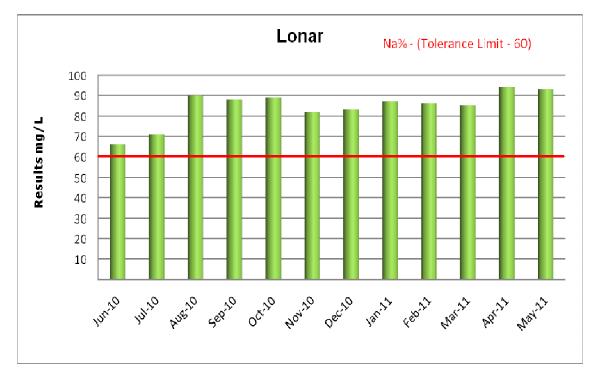
Graphical Representation of Chloride (As per ICAR Standards Parameter) for the Year 2010-2011 WR Division Aurangabad Lonar Station



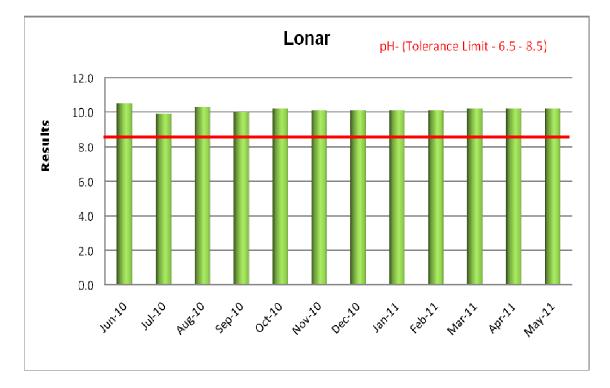


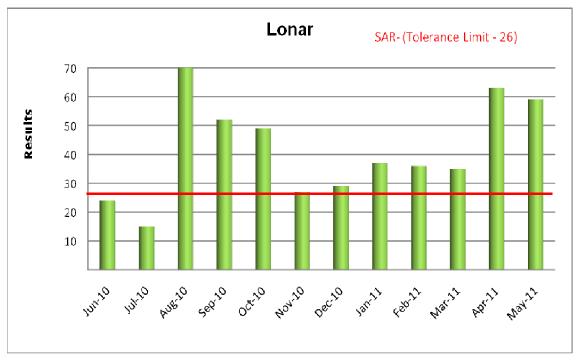
Graphical Representation of Chloride (As per ICAR Standards Parameter) for the Year 2010-2011 WR Division Aurangabad Lonar Station



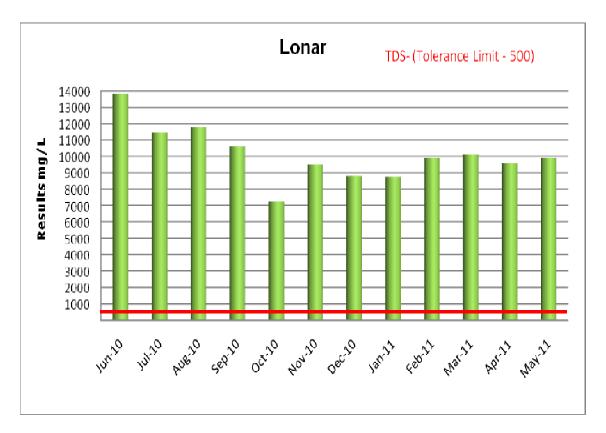


Graphical Representation of Electrical Conductivity (As per ICAR Standards Parameter) for the Year 2010-2011 WR Division Aurangabad Lonar Station





Graphical Representation of Na% (As per ICAR Standards Parameter) for the Year 2010-2011 WR Division Aurangabad Lonar Station



LONAR



LONAR Famous for : In the whole world, this is the biggest natural saltwater lake.

The shape of this lake is like a bowl. This lake came into existence when a huge meteor from the space descended on the earth some 50,000 years ago. This lake is 7 km in consideration and the diameter is 1.8km. This is the oldest and the largest meteoric crater in the whole world.

This lake was created by the hypervelocity of meteoritic affect in basaltic rock and the research has shown that the meteor, which had stroked was sixty meters in diameters.Lonar has been recorded to be the third biggest natural salt-water lake in the world.

One of the British officers C.J.E Alexander noticed this lake and in 1896, GK Gilbert conducted thorough research to prove the existence. The lake was proved to be formed due to the meteoric hit by using Cosmic Ray.

This crater is gifted with distinctive self-emerged ecosystem heading to evolution of new forms of life. Blue green algae, Spirula are found in the lake. This lake is an attractive brackish lake. There are many features of moon rocks found in the rocks here.

This lake has been mentioned in the ancient scripts. It is mentioned in Aain-i-Akbari .There was a salt factory during the era of Emperor Akbar.The name of the lake has been mentioned in Sanskrit literature Viraj Mahatma. In ancient time, this lake was known, as Viraj Kshetra.There is an old temple of Lord Vishnu, which was built in the 13th century during the era of Chalukya. This temple has a width of (450m) and is (150m) long .This temple is a magnificent example of huge and lustrous ancient structure.

The Paphareshwar pilgrimage is close to this lake .This temple has ancient Sanskrit engravings, which are still not deciphered.

CHAPTER – V RESULT & DISCUSSION

CHAPTER - V

CONCLUSION

Designated best use	Quality Class	Primary Water Quality Criteria
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5.0 Classes of Water:

Source: 2295, 1982 & (IS10500, 1983)

Class A:

Surface water for use as "Drinking water" Sources without Conventional Treatment

Class B:

Surface water for out door bathing

Class C:

Surface water used as "Drinking Water" Source within conventional treatment followed by disinfection.

Class D:

Surface water used for fish culture & wild life propagation.

Class E:

Surface water for irrigation, industrial cooling, waste disposal.

Drinking water source	А	> Total coliform organisms (MPN/100 ml) shall be	
without conventional		50 or less	
treatment but with		▶ pH between 6.5 and 8.5	
chlorination		➢ Dissolved Oxygen 6 mg/l or more, and	
		Biochemical Oxygen Demand 2 mg/l or less	
Outdoor bathing	В	> Total coliform organisms(MPN/100 ml) shall be	
(organized)		500 or less	
_		▶ pH between 6.5 and 8.5	
		Dissolved Oxygen 5 mg/l or more, and	
		Biochemical Oxygen Demand 3 mg/l or less	
Drinking water source	С	> Total coliform organisms(MPN/100 ml) shall be	
with conventional		5000 or less	
treatment		➢ pH between 6 and 9	
		Dissolved Oxygen 4 mg/l or more, and	
		Biochemical Oxygen Demand 3 mg/l or less	
Propagation of wildlife	D	> pH between 6.5 and 8.5	
and fisheries		Dissolved Oxygen 4 mg/l or more, and	
		➢ Free ammonia (as N) 1.2 mg/l or less	
Irrigation, industrial	Е	> pH between 6.0 and 8.5	
cooling, and controlled		≻ Electrical conductivity less than 2250 micro	
disposal		mhos/cm,	
		Sodium Aborption Ratio less than 26,	
		\geq and Boron less than 2 mg/l.	
	Below E	➢ Not Meeting A, B, C, D & E Criteria	
CPCB Water Quality Criteria			

CPCB Water Quality Criteria

5.1 Surface water:

In order to understand drinking water contamination, it is necessary to first understand from where our drinking water comes. For most urban residents, relying upon municipal water systems, drinking water comes from two major sources, groundwater and surface water. These two sources of drinking water will be referenced throughout this guide to water contamination.

Surface water refers to water occurring in lakes, rivers, streams or other fresh water sources used for drinking water supplies. While most drinking water is withdrawn from groundwater sources, surface water remains a significant water resource.

5.2 Contaminants resulting from Biological Matter in Water -Human and Animal Feces:

Though there are multiple ways that pathogens and harmful microorganisms enter the water supply, the principal means of entry is through water contamination by human sewage and/or animal feces. These types of contamination largely affect surface water areas like rivers, lakes and streams from which drinking water is taken. Though most of the diseases caused by drinking water containing fecal material are gastrointestinal, such water can cause more serious and life threatening diseases like hepatitis (A, B and C) and Legionnaire's disease. Hepatitis is an inflammation of the liver, characterized by jaundice, fever and abdominal pain. Acute cases of hepatitis, especially hepatitis C can be fatal. Legionnaire's disease is an infectious, sometimes fatal disease that is characterized by high fever, incessant cough, lung congestion and subsequent pneumonia. The disease can permanently damage such vital, internal organs as the heart and lungs. Ingestion of drinking water contaminated by human or animal feces can also result in higher rates of spontaneous abortion in pregnant women. The effects of pathogencontaminated drinking water are especially detrimental to cancer patients, as well.

Human and animal feces enter water systems primarily through breakdowns in sewage and improper treatment of surface water sources. Human and animal fecal matter affects surface water almost exclusively, but as surface water bleeds into groundwater through stream and riverbeds, the groundwater can become contaminated as well. Water treatment facilities can certainly lower the presence of human and animal feces in drinking water but they cannot remove it entirely. Current treatment methods are aimed at protecting surface water sources from contamination before the water enters a treatment plant.

5.3 Significance of inorganic chemicals in drinking water:

Chloride: The secondary drinking water standard for chloride is 250 mg/L, because some people can detect a salty taste when chloride exceeds 250 mg/L. Chloride itself has no health effect. However, the amount of chloride can be directly tied to the amount of sodium in the water. To find the sodium level of this sample, multiply the chloride level by 1.6. Those individuals, who have high blood pressure and monitor their salt intake, may want to let their physician know if the level is over 250 mg/L.

Fluoride: The primary drinking water standard for fluoride is 4 mg/L, and the secondary standard is 2 mg/L. A fluoride concentration of approximately 1.0 mg/L helps to prevent dental cavities and osteoporosis. At concentrations above 2.0 mg/L, fluoride may cause mottling of enamel of permanent teeth (most common in children up to age 10 while teeth are forming). Bone changes can occur if drinking water contains more than 4 mg/L fluoride and if fluoride exceeds 20 mg/L, crippling fluorosis can occur after long term consumption.

Nitrates: The primary drinking water standard for nitrate (as N) is 10 mg/L. Excessive nitrate consumption by infants less than one year of age

may result in "blue baby" syndrome, also known as methemoglobinemia. Nitrate replaces the oxygen in red blood cells causing an oxygen deficiency in the infants. At the extreme, it can be fatal. High nitrates also effect ruminants (sheep, cattle, etc.) Which are sensitive to nitrates. Elevated nitrates are thought to not pose a direct health threat to children over the age of one or adults

pH: The pH scale extends from 0 (which is very acidic) to 14 (which is very alkaline), with 7 being neutral. Drinking water should ideally range from 6.6 to 8.5. Lower pH tends to make metals and hardness minerals more soluble, possible allowing unwanted heavy metals into a water supply. pH levels below 6.0 or above 10.0 could also indicate the presence of a contamination source

Total dissolved solids (TDS): TDS is a measure of all dissolved inorganic material in water. TDS over 1,000 mg/L is objectionable because of the mineral taste. Harm to humans has not been proven. However high TDS (over 400 mg/L) does shorten the lives of water heaters. Concentrations more than 1,000 mg/L can accelerate corrosion in general.

Total Hardness: Also called "hardness as Calcium Carbonate." Calcium and magnesium are the principle minerals contributing to Total hardness; coming from soil and rocks where water readily dissolves them. Iron and manganese can also contribute to the Total Hardness. Water with less than 50 mg/L is considered soft. Although hard water requires more soap and detergent for laundering and deposits scale on fixtures, soft water may be corrosive. Hard water may also possibly aid in the prevention of heart and arterial diseases. Even so, a Total Hardness greater than 400 mg/L is considered excessive hardness as calcium: Hardness as Calcium is usually compared to Total Hardness. When the amount of Hardness as

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Calcium is subtracted from the Total Hardness reading, the difference is the amount of magnesium in the water. Calcium is usually more prevalent than magnesium. Excessive calcium has been implicated in formation of kidney or bladder stones, while high levels of magnesium may have a laxative effect on these not used to high levels.

Explanation of bacterial analysis:

Bacteria: At the time of testing for bacteria two results are given – total coliform E.coli. The first, total coliform, is a test for coliform bacteria which are distributed widely in the environment. They are on animals, plants and in the solid, but are in large numbers in the feces of warmblooded animals. A positive reading for total coliform bacteria means that the water supply has been affected by the environment, and disease-causing organism may or may not be present. However, it is cause for concern and corrective action, such as well chlorination, should be taken. E.coli is a species of bacteria found in the intestinal tract of warm-blooded animals. When a sample is positive for E.coli, it means there is fecal contamination in the well water. This could be human or animal, but its implications are more serious than total coliform. The well's construction should be reviewed and the well chlorinated, with a follow-up test done 10 days after chlorination. A negative reading means none of the above bacteria were found in the sample.

CONCLUSIONS

5.4 PART I: Conclusion for River Samples:

In the Year 2010-11 it can be concluded that for the Aurangabad Division Station & Amaravati Division Stations water is good for irrigation purpose without any treatment but it is observed that the stations like Aurangabad & Lonar, the contamination is very high. This is mainly due to Contamination of Industrial waste & Sewage waste. Hence, it can be concluded that water from all these sources is not suitable for Irrigation purpose; it requires treatment before its usage.

5.5 Part II: Conclusion for Dam Samples:

The threat of harmful contaminants in drinking water can no longer be reasonably ignored. The correlation between contaminated drinking water and many significant diseases and health problems is far too strong to discount.

Of course, municipal water treatment facilities have lowered the presence of many of the more harmful contaminants and it has set maximum contaminant levels below which it is assumed that contaminants may be safely ingested into the body. Municipal treatments facilities are not infallible and its levels do not represent a safety level for every person. Children, the elderly and those individuals who already have weakened immune systems, are particularly at risk to drinking water contaminants. Two of the most volatile drinking water contaminants chlorine and fluoride are actually treatment additives. Also lead is another more harmful contaminants, enters drinking water after treatment and cannot be regulated by municipal water systems. Therefore, municipal water systems cannot and should not be trusted to provide healthy, clean drinking water.

There are many home treatment alternatives that can purify drinking water to a greater extent than city treatment plants. Reverse osmosis and distillation two of these alternatives are moderately successful at removing some contaminants, but they are expensive and wasteful. Bottled water besides being expensive and highly unfeasible as a main drinking water source is not under the same government regulations as municipal water systems and may actually contain more contaminants than tap water. The absolute best technology now available for treating water and removing undesirable contaminants is water filtration. Water filters when compared to any other water treatment alternative will remove more contaminants and provide safer, healthier drinking water.

5.6 REMEDIAL MESAURES:

- Use of such water for salt tolerance crop is recommended based on special study.
- Before letting out to the down stream reservoir such source of water to be taken into consideration for the specific use.
- Classification of source may be as per use of water for irrigation based on Sodium Absorption Ratio, Percent Sodium, and Residual Sodium Carbonate.
- Effluent from non point sources to be treated before discharging into the river, Use of direct source of water to be avoided

CHAPTER – VI OTHER ACTIVITIES

CHAPTER-VI

OTHER ACTIVITIES

6.1 REVENUE GENERATION TO GOVERNMENT OF MAHARASHTRA

Apart from monitoring of water quality network for Water Quality lab level II at Aurangabad, the infrastructure facility is made available to the users from various Government, Non Government, Private sector as well as individuals.

The facility is availed by many users with testing of sample towards drinking purpose, ice factory, construction purpose, swimming tanks, irrigation purpose & study purpose.

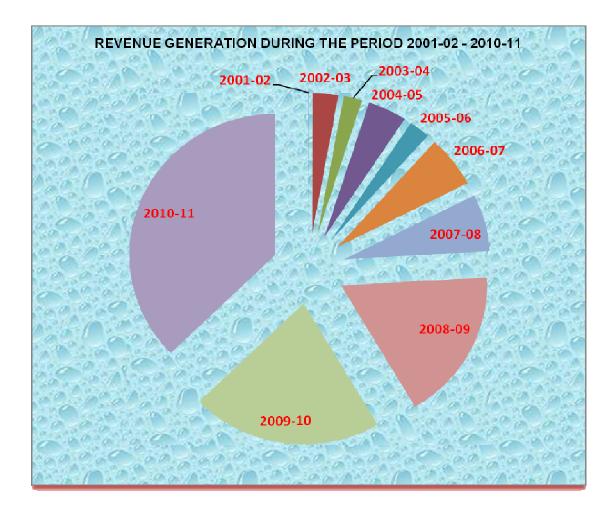
During the year 2010-11 many clients approached to the laboratory. The valuable clients availed the facility of the laboratory are as below;

Revenue Received from year 2001-02 to 2010-11

Year	Aurangabad Lab
2001-02	408
2002-03	35,995
2003-04	25,941
2004-05	54,248
2005-06	29,557
2006-07	74,652
2007-08	82,879
2008-09	2,20,243
2009-10	2,67,252
2010-11	4,70,278



Pie Diagram Of Revenue Received from month year 2001-02 to 2010-11 Water Quality Lab Level II Aurangabad



Annual Report

Water Quality Monitoring through Water Quality Lab Level-II Aurangabad

for the Year 2010-11

ANNEXURE

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List of Selected Clients OF 2010-11 Analyzed water Sample

	Analyzed water Sample	_
Sr. No.	Name of Clients	Purpose
1.	Chief Officer Muncipal Council Sindhkhed	
	Raja	
	M.A.E.T. Hotel Mangement	
	M.A.E.T. Y.B. Chauhan college of	
	pharmacy	
	M.A.E.T. Millinium Institute	
	M.A.E.T. tom patric inst of computer	
	Sigma Institute of medical Scien Pvt Ltd	
	Good Year SouthAsia Tyers ,A'bad	
	Bharkha Farm Paithan	
	MGM Polytechnic College	
	Neel Operations, Abad	
	Godavari jaldhara,A'bad	
	Airport Authority of India, Aurngabad	
	Bhaskaracharya Pratishthan,Chate	
	English school A'bad	
	Auranhgabad Jalna Tollways Ltd.A'bad	
	Gram Aryogya Pani Purwatha	
	OMR Bagla	
	Franky India Pvt Ltd.Abad	
	KT Sangam Infrastucture	
	Regional Manager, Frozen semen	
	Laboratory, Harsul, Aurangabad	
	Marathwada Urban Banks Co Op Aso, Ltd	
	A'bad	
	Welcom Hotel Ram International A'bad	
	Mukteshwar Suger Mill	
	Orchid Pharma Ltd. Waluj A'bad	
	Saylee Trust College of	
	Pharmacy, Mitmita, Aurangabad	
	Lemon Tree Hotel A'bad	
	Nirmal Water System	
	Muncipal Corporation Aurangabad	
	Gram Arogya Water Supply & Sanitation	
	Comitte	
	Purti Power & Sugar Ltd.Nagpur	
	Foster Development School of	
	Management A'bad	
	OMR Bagla Automotive A'bad	
	ndian Oil Co. Ltd.	
	Dr. Babasaheb Ambedkar Marathwada	
	University A'bad	
	MIDC E & M Division A'bad	
	Ecosterile Marketing Pvt.Ltd.Mumbai	

	Hotel Trimurthy	
	Depo Manager MSRTC Soygaon	
	Junior Telecom Officer BSNL Civil Sub-	
	Division Shrirampur	
	Glanze Engg. Co. Ltd.	
	Executive Engineer, Majalgaon Irrigation	
	Divison,Parli	
	Saptshrungi steel Pant,Jalna	
	M/s.Beten Concrete Products	
	M/S.Tulsi Buildcon,A'bad	
	Hotel Abhi,A'bad	
	Peri Fresh Builtin ,A'bad	
	New Arts ,Science,Commerce	
	College,Shevgaon Dist.Ahmadnagar	
	Jaylaxmi casting ,Farola	
	ILFS Water Limited A'bad	
	Om Sai Industries,A'abad	
	Ajeet seeds Ltd,Chitegaon Tg.Paithan	
	dist A'bad	
	Bohra Foods Pvt.Ltd.Jamner Dist Jalgaon	
	Arti Farms Pvt.Ltd	
	Mukteshwar Sugar Mills Ltd	
	Endurance Technology Ltd	
	Navkisan Bioplant Tech Ltd.	
	Grampanchayat Office, Krishnapurwadi,	
	Sawngi	
	Dhoot Transmission Ltd. A'bad	
	Shakuntala Industries Gangapur	
	Kashvi Chemicals, A'bad	
	Deputy Engg. MIDC, E & M, Sub Division	
	Aurangabad	
	Videocon Ltd. A'bad	
	Village Health Water & Sanitation	
	Comm. Janjala, A'bad	
	Yash Assocites abad	
	Jailakshami Casting Alloy	
	Dy Engineer MIDC EM sub division Abad	
	Executive Engineer MIDC Abad	
	Asst Engr sholapur CentralSub Divn	
	Indian Institute of rural Development	
	Lemontree Hotel abad	
	Siddhivinayak Enterprizes Abad	
	Fortrees Infrastructure A'bad	
	Group Grampanchayat Paithnkheda	
	Bembde Hospital A'bad	
	SVS Chemical Corporation Pune	
	M. G. M. J. N. E. College	
·		L

Bajaj Finance	
Hindalco Almex Aerospace	
 Pantallons Retail Prozon Mall	
Savera Auto Comps	
Larsen Toubro Itd ecc divn	-
Water life Kranti Chowk ABAD	
Principal Dr Zakria institute of tech	
 science	
RADICO NV Distillaries, Shendra, A'bad	
Sectional Engineer,South Central railway, Jalna	
Principal, New Arts ,Science College,	
Shevgaon Dist Ahmednagar	
Nath Polytechnic, MIDC, Paithan	
Rock & Water Scape International Super	
stone KoatsPvt Ltd.Joint	
Venture,Sakinaka,Mumbai	
Shree Swami Samarth	
Industries, Chikalthana, A'bad	
Kaygaon Paper mill	
More Mega Store A'bad	
Excedy India Ltd. A'bad	
Aurangabad Electricals Ltd., A'bad	
WIPRO ltd.Consumer Care & Lightening	
Dn.A'bad	
Ion Exchange India Ltd. Dapoli Pune	
Dhoot Trasmission Pvt. Ltd. Aurangabad	
SE, Signal (SE Railway) Nanaded	
Kamalnayan Bajaj Hospital	

QUALITY POLICY

WATER QUALITY LABORATORY, LEVEL-*I*I, AURANGABAD. DECLARED THE PURPOSE OF THE ORGANIZATION i.e

COLLECTION & TESTING OF WATER SAMPLES

HAS LAID DOWN THE FOLLOWING QUALITY POLICY, THAT HAS BEEN COMMUNICATED AND UNDERSTOOD WITHIN THE ORGANIZATION AND HAS PROVIDED ADEQUATE FRAME WORK FOR REVIEWING ITS QUALITY OBJECTIVES AND QUALITY POLICY FOR CONTINUING SUITABILITY AND IS

COMMITTED.....

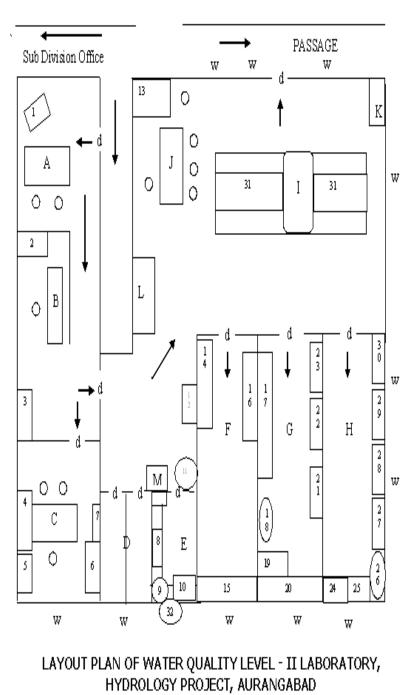
> TO MONITOR EFFECTIVENESS OF QMS TIME TO TIME & WILL ALSO WORK FOR CONTINUAL IMPROVEMENT OF THE ACTIVITIES.

> TO WORK FOR CONTINUAL IMPROVEMENT IN ITS TECHNOLOGY, PROCESSES AND TO INCREASE COMPETENCY LEVELS HAVE ITS PERSONNEL.

> TO DELIGHT THE CUSTOMER BY FULFILLING CUSTOMER NEEDS, STATUTORY/REGULATORY REQUIREMENTS AND ANY REQUIREMENT WHICH IS NOT STATED BUT WHICH IS REQUIRED FOR APPLICATION OF CUSTOMER SERVICE.

Rev. No.: 00 DATE: 02 March 2009. EXECUTIVE ENGINEER HYDROLOGY PROJECT DIVISION, AURANGABAD

LEGEND S



1	Display Board
2	Cup Board 1
3	Cup Board 2
4	Computer Table
5	Inverter
6	Cup Board 3
7	Cup Board 4
8	Distilled Water Assembly
9	Sink
10	Kjaldahl Distillation Unit
11	Sink
12	Refrigerator
13	Cupboard Lab 1
14	Non Hazardous Chemicals
15	Desigator, Weighing Balance
16	Hazardous Chemicals
17	Working Table
18	Autoclave
19	Distillation Unit
20	Centrifuge & Water Bath Unit
21	BOD Incubator
22	Deep Fridge
23	Bacteriological Incubator
24	Computer
25	Spectrp Photometer
26	Sink
27	Turbidity Meter
28	pH Meter
29	EC Meter
30	COD Digester Unit
31	Analysis Platform
32	Exhaust
A	Reception Table
В	INWARD & OUTWORD
С	Lab Office
D	Toilet
E	Distillation Unit/Pantry
F	Chemical Storage Room
G	Microbiological Room
Н	Instrumental Room
Ι	Washing Bath
J	Chemist Table
К	Waste Disposal Area
L	Received Sample Location
М	Field Kit Cupboard
	· · · · · · · · · · · · · · · · · · ·

NEWS & PEPER CUTTING



'चपटी'चा वापर करीत असुल्याने नागरिकांच्या आरोग्याचा 🔳 पाणी तपासणीचा 'खेळखडोबा' 🗏 शासनच करते नियमाचे उल्लघन

झाकणावर कागद गुंडाळला जातो किंवा वेष्टन लावले जाते. मात्र, येथे फेकन दिलल्या बाटल्या

भिगरी, संत्रा, नारंगी या देशी दारूच्या बाटलीस 'चपटी' म्हटले जाते. सर्वत्र सहज उपलब्ध होणारी ही चपटी कचऱ्यात फेकून दिली जाते. कचरा वेचणारे या बाटल्या जमा करून भंगारात विकतात. भंगारातील या बाटल्या प्रादेशिक आरोग्य प्रयोगशाळा खरेदी करते. या बाटल्या नंतर पाण्याचे नमुने घेण्यासाठी वापरल्या जातात. मध्यंतरी जमा झालेल्पा क्विंटलभर दारूच्या रिकाम्या बाटल्या प्रादेशिक आरोग्य प्रयोगशाळेने विकून टाकल्पा, अश्री माहिती कर्मचान्यांनी दिली.

झाकणावर प्रश्नाचन्ह

प्रादेशिक आरोग्य प्रयोगशाळेत बाटली निर्जंतुक करण्यासाठी हॉट एअर ओव्हनमध्ये २ तास ठेवली जाते. या देशी दारूच्या बाटल्पांवरील झाकण कृचकामी असते. या तापमानात ते तग धरू शकेल काय, असा प्रश्न निर्माण होतो. झाकण तुटल्पास पाण्याचे नमुने गोळा करणारे कर्मचारी दसऱ्या निर्जतक केलेल्या बाटलीचे झाकण लावतात. मग पाण्याच्या गुणवत्तेवर याचा परिणाम होतो की नाही, या प्रश्नावर अधिकाऱ्यांनी उत्तर दिले नाही.

तपासणाच नियम

- पाण्याची सूक्ष्मजीवशास्त्रीय तपासणी करण्याकरिता काचेचे घट्ट बसणारे अथवा धातूचे फिरकीचे झाकण असलेली २०० मिलि ची बाटली असावी, ही बाटली पाण्याचा नमुना घेण्यापूर्वी निर्जतुक असणे आवश्यक आहे. ही बाटली पाणी नमुना घेतेवेळीच उघडावी. 📕 झाकणाच्या आतील बाजूस स्पर्श न करता बाटलीचे झाकण उघडावे. 📕 बाटलीत पाणी घेतल्पावर झाकण पूर्ववत लावून पुन्हा उघडू
- नये. यासाठी झाकणावर कागदी आच्छादन लावावे. 🛚 बाटलीवर नमुना घेतलेल्पा ठिकाणचे नाव, पत्ता लिहिणे महत्त्वाचे

नळाच्या पाण्याचा नमुना घेण्याची पद्धत 🛚 सुरुवातीस नळाची तोटी बाहेरून धुऊन काढावी.



औरंगाबाद, दि. २३- प्रादेशिक आरोग्य प्रयोगशाळेत चव देशी दारूच्या बाटलीत आणलेले पाण्याचे नमुने तपासले ज आहेत. यामुळे पाण्याची सुक्ष्मजीवशास्त्रीय तपासणी कित गुणवत्तापूर्ण होते यावरच प्रश्नचिन्ह निर्माण झाले आहे. मनपाचा पाणीपरवठा विभाग असो की जिल्हा परिषदेचे प्राथमिक आरोग्य केंद्र येथे पाण्याच्या नमुन्यासाठी सर्रास या सरकारी यंत्रणेने खेळ मांडला आहे.

सार्वजनिक पाणीपुरवठा, विहीर, बोअरचे पाणी पिण्यासाठी योग्य एकाही बाटलीच्या झाकणाची दक्षता घेतली गेल्याचे दिसून आले नाही. का? पाणी दूषित होऊन साथीचे आजार पसरू नयेत किंवा त्वचारोग । देशी दारूच्या बाटतीवरीत झाकण एवढे कुचकामी असते की ते नयेत यासाठी शासकीय यंत्रणा नळाचे पाणी, विहीर, बोअरमधील पाण दाबत्यास लगेव तुदते. मग या पाण्याच्या नमुन्यात हवा न शिरण्याची नमुने तपासणीसाठी वेळवेळे घेत असते. या पाण्याच्या नमुन्याची तपाः काय गॅरंटी, असा प्रश्नही उपस्थित होतो. चप्पल-बूट घालूनच सर्व छावणी येथील प्रादेशिक आरोग्य प्रयोगशाळेत करण्यातः ज्या प्रयोगशाळेत वावरतात. प्रयोगशाळेवे सर्व दरवाचे-खिडक्या महानगरपालिका हद्दीतील विविध वसाहतींमधील पाण्याचे दररोज १० = सतत उपड्या ठेवल्या जातात. प्रयोगशाळेत धूळ नसावी यासाठी तपासणीसाठी घेतले जातात. मनपा पाणीपुरवठा विभाग व आरोग्य प्रयोगशाळेचा आतील भाग बंदिस्त असावा. या प्रयोगशाळेत पाण्याचे नमुने घेते. याशिवाय १३ ग्रामीण रुग्णालये व ५० प्राथमिक आः दताखान्यतील ऑपरेशन थिएटरसारखी काळजी घेतली केंद्रांतून पाण्याचे नमुने तपासणीसाठी आणले जातात. पाण्याचे नमुने घेत पाहिजे. हे नियम येथील प्रादेशिक आरोग्य कोणती काळजी घ्यावी, पाण्याचे नमुने कोणत्या बाटलीत आणावेत, प्रयोगशाळेला लाग्व होत नाहीत, असे येथे पाहणी

राज्यसेरात त्याच बाटल्या

प्रादेशिक आरोग्य प्रयोगशाळेतील मुख्य अणुजीवशास्त्रज्ञ सु.वि. पत्की यांनी सांगितले की, अनेक वर्षांपासून प्रयोगशाळेत दारूच्या बाटल्यां पिण्याच्या पाण्याचे नमुने घेण्यासाठी वापर केला जातो. या बाटल्यांचे वॉशिंग केले जाते, त्यानंतर निर्जंतुकीकरण्यासाठी हॉट एअर

ओव्हनमध्ये १६० डिग्री सेंटिग्रेड या तापमानात २ तास ठेवले जाते. या निर्जंतुक केलेल्या बाटल्याच मनपा, ग्रामीण रुग्णालये, प्राथमिक आरोग्य केंद्रांना पाण्याच्या नमन्यांसाठी दिल्या जातात. मध्यंतरी आम दारूच्या बाटल्यांवरील झाकण बदलण्याचा प्रयत्न केलाः पण तो प्रयो अयशस्वी ठरला. या दारूच्या बाटल्यांत पाण्याचे नमुने घेण्याबाबत आजपर्यंत कोणी आक्षेप नोंदविला नसल्याने या बाटल्पांचाच सर्व आरोग्य प्रयोगशाळेत वापर होत असल्याचे त्यांनी नमुद केले.

बाटलीस कोणते झाकण लावावे, त्या बाटलीत हवा जाऊ नये यास नियमानुसार लांब-गोल आकाराच्या कोणती खबरदारी घ्यावी यासाठी कडक नियम आहेत; पण हे सर्व नि बाटतीचा वापर केला जात आहे. या धाब्यावर बसवून पाण्याचे नमुने घेतले जात आहेत. दारूच्या फेकून दिले बाटतीच्या झाकणाला आतून आटे असून, ते बाटल्यांचा पाण्याचे नमुने घेण्यासाठी सर्रास वापर केला जात अ बाटलीस घट्ट बसते. त्यावर ॲल्युमिनियमचा उल्लेखनीय म्हणजे, पाण्याच्या नमुन्यासाठी शासनच दारूच्या बाटल्य कागद लावून खराने बांधते जाते. यामुळे पुरवठा करीत आहे. देशी दारूची बाटली चपटी असते व ने-आण करण्य पाण्याचा नमुना असतेत्या बाटतीत हवाच सोयीस्कर असते. एका बॅगमध्ये दहा ते पंधरा बाटल्या आरामशीर मावत जाऊ शकत नाही प्रयोगशाळेत या

केल्यावर दिसून आले.

दसरीकडे राज्य सरकारच्या जलसंपदा

विभागांतर्गत जलविज्ञान प्रकल्प जागतिक बँकेच्या

सहकायनि जलगुणवत्ता प्रयोगश्चाळा स्तर-२ येथेही

पाण्याचे नमुने तपासले जातात. डॉ. हेडगेवार

हॉस्पिटलच्या बाजूला ही शासनाची प्रयोगशाळा आहे.

फेब्रुवारी महिन्यातच या प्रयोगशाळेस आय.एस.ओ.

९००१-२००८ हे प्रतिष्ठेचे मानांकन प्राप्त

झाले. या प्रयोगशाळेत आणि प्रादेशिक आरोग्य

प्रयोगशाळेत मोठी तफावत दिसून आली. एका

ंशासकीय प्रयोगशाळेत पाण्याच्या

नमुन्यांसाठी देशी दारूच्या बाटलीचा वापर

होत आहे, तिथे दुसऱ्या शासकीय

प्रयोगशाळेत खास पाण्याच्या नमुन्यांसाठी जागतिक आरोग्य संघटनेने ठरवून दिलेल्या

Festival Shoppee gets underway

Aurangabad, Oct 28: The five-day exhibition-cum-sale 'Festival Shoppee' began at S F S High School ground on Thursday.

The leading industrialist Yashwant Bhogale inaugurated the exhibition.

Commander Anil Save, Ram Bhogale, Madhav Bhogale, Anjali Save, Rajni Bhogale, Shyam Bhogale, Savitri Bafna, Ashish Garde and others were also present.

More than 100 companies have set up its stalls of the branded goods in the exhibition. These include Hyper, Godrej, Onida, BSA, Nirlep, Prestige, Pigeon, International, Usha Filltronics, Mitashi, Fuji Kodak, Bajaj, Film. Maharaja, Camlin, Ganesh Papad, Pitambari, Gala Brush, Prince Plastic, Wagh



Industrialist Yashwant Bhogale inaugurating the 'Festival Shoppee' exhibition at S F S High School ground on Thursday. Also seen are Cdr Anil Save, Ram Bhogale, Madhav Bhogale and Anjali Save.

Sudarshan Saur.

items like 'bhail puri, vada

Bakri Chaha, Pravin Masile, pav, Chinese vegetarian and Ram Bandhu, Tata B P Solar, non-vegetarian, juice and Cycle Agarbatti and gobi manchurian'. The entry is free and the exhibition will There are also stalls of food remain open from 11 am to 9 pm till November 1. -(LNS)

-2-ABONNET BR

संभाजीनगर, शुक्रवार, दि. २० ऑगस्ट २०१० शवदच पान



अनिल भाले

<u>आनल भाल</u> संभाजीनगर, दि. १९ - शहरातील बहुतांशी वसाहर्तीमध्ये नागरिक बोअरच्या पाण्यावर तहान भागवत आहेत. दरम्यान, शहरातील काही खासगी व संस्थांच्या बोअरच्या पाण्याची तपासणी केली अराता ९० टक्के नमुन्चांत झारांचे प्रमाण आवश्यकतेपेक्षा जास्त असल्याचे जलसंपदा विभागाच्या जलगुणवत्ता प्रयोगशाळेने केलेल्या वापाणील आवलन आजे आहे स्वरक्षत द्वीव विभागीच्या जलगुणवत्ता प्रयोगशाळेन केलेल्या तपासणीत आढळून आले आहे. सारयुक्त व दूषित पाण्यामुळे जडणाऱ्या व्याधींचे प्रमाण वाढत आहे. शहराचा सर्व बाजूनी मोठ्या प्रमाणात विस्तार होत असुन, अनेक वसाहतींमधील नागरिक व खासगी अस्थापने, शहरानजिकच्या हॉटेल्स,

हॉस्पिटल्सना बोअरच्या पाण्यावरच आपली तहान भागवावी लागते. अलीकडे जलजन्य आजारांच्या रुग्णांत मोठ्या प्रमाणात वाढ झाली आहे. क्षारयुक्त उण्णात माञ्या प्रभागत वाढ झाला आह. सारयुक्त व दूषित पाणीच विविध आजारास कारणीभूत उस्ते. त्यामुळे पोटांचे विकार, कावीळ, संसर्गजन्य रोग, दंत क्रिडनीचे विकार त्वचारोग, कावीळ हे आजार जडतात. जलसंपद्ा विमाग जलविज्ञान प्रकल्पाच्या जलगुणवत्ता प्रयोगशाळा स्तर- २ मध्ये जून २००९ ते मे २०१० या वर्षांत ४९० पाण्याचे नमुने तपासण्यात आले. यातील तपासणी बोअरच्या ९० टक्के नमुन्यात क्षारांचे प्रमाण आवश्यकतेपेक्षा जास्त आढळून आले आहे. या



हां प्रथागशाळो अत्याधानक उपकरणाना सज्ज असून, येथील कर्माचारी उच्चाशीक्षत आहेत. प्रयोगशाळेत मराठवाडवातील मोठी घरणे, नद्यांचे पाणी नमुने संकलित करून त्याची भौतिक, रासायनिक, जैविक तपासणी करून त्याचे निष्कर्ष संगणकी प्रणालीद्वार जतन केले जातात. या प्रयोगशाळेस फेब्रुवारी २०१० मध्ये आय एस. औ. १००१२ - २००८ हे प्रतिष्ठेवे मानॉकत प्राप्त झाले आहे. प्रयोगशाळेमार्फत जलसाक्षरतेच्या प्रसारासाठी विविध उपक्रमांचे आयोजन करण्यात येते.

प्रयोगशाळेत आतापर्यंत २०० शासकीय तर अयागशाळत आतापयत २०० शासकाय तर १४०० नमुने हे खासगी संस्था, व्यक्ती, कारखाने, पंचतारांकित हंटिल्स, हॉस्पिटल्स असे एकूण १६०० पाण्याचे नमुने तपासण्यात आले आहेत. बोअरच्या पाण्याच्या तपासणीत क्षाराचे आहेत. बोअरच्या पाण्याच्या तपासणीत क्षाराचे प्रमाण प्रतिचिटर १००० मिलिग्राम ते १२०० मिलिग्रॅम एवढे जास्त प्रमाणात आढळून आले आहे. प्रहिति ग्रॅमएवढेच असणे साधारणतः ५०० मिति. ग्रॅमएवढेच असणे मि. ति. ग्रॅमएवक असावा. मात्र हे प्रमाणा ५०० ते ७०० मि. ती. ग्रॅमपर्यंत गेल्याचे तपासणीअंती आढळून आले आहे. जलगुणवत्ता तपासणी आवश्यक या संदर्भात जलविज्ञान प्रकल्पांचे उपविभागीय अधिकागी वरी पी कलक्ताणों की प्रयोक

अधिकारी व्ही. पी. कुलकर्णी म्हणाले की, प्रत्येक नागरिकाने वर्षातून २ वेळा आपण पितो त्या पाण्याची गुणवत्ता तपासणे आवश्यक आहे. दूषित पाण्यामुळे जडणाऱ्या रोगांच्या उपचारावरील खर्च पाणी तपासणीच्या खचपिक्षा अधिक असतो प्रत्येकाने पाण्याची तपासणी करणे आवश्यक आहे. बोअरच्या पाण्याची जलगुणवत्ता आवश्यक असून, क्षारांचे प्रमाण कमी करण्यासाठी वॉटर क्युरी बसवणे आवश्यक असल्याचे ए. डी. महाजन यांनी सांगितले

अत्याधुनिक प्रयोगशाळा

राज्य संरकारच्या जलसंपदा विभागाअंतर्गत जलविज्ञान प्रकल्प जागतिक बैकेच्या सहकायनि १९१५ पासून सुरू करण्यात आला. या प्रकल्पांतर्गत राहेरातील हैडगेवार हॉस्पिटलनजीक २००१ पासून जलगुणवत्ता प्रयोगशाळा स्तर - २ कार्यान्वित करण्यात आली. ही प्रयोगशाळा अत्याधुनिक उपकरणांनी

Visitors visit to the Laboratory and their remarks

Visitor's Book				
DATE	NAME & ADDRESS	CONTACTS	E-MAIL	REMARKS
9/08/90	औ. राजन्द्र वन्सीलात ज्युक्ला.	0280-	shuklarqi 2001@ yahoo.	प्रयोगगाळा अत्युत्वृष्टु आहे.
etc:	मुरव्य आभियंता. महाराष्ट्र जलसंपत्ती विन्तास	2368143	(0·1)	150 मिनाव्यामुळे एक मानाचा
1	महाराष्ट्र जारमावार.			तुरा रवीवता गैला आहे. सामान्य जनमानसात शुणवत्त्रेबाबत जनजाम्
				आवणे गरजेये आहे जजुन प्रयत्न
218/10				केलं तर भाविष्यात अरपूर नाम
which !	(A			होरे रामेल . प्रसारमाह्यमातून
前的	85			सरप्र प्रासिष्ठी स्वावी ही सूचना,
NUP70	San I. C.D.			R85lunder

	1	lisitor's	Baak	***
DATE	NAME & ADDRESS	CONTACTS	E-MAIL	REMARKS
25-6-1D	T.S.R. Gupte DAG 0/0 A.G (AU) II Nagpar	9422305448	Esogupta & gmail. com	Lab is one of the best and observing will lae of down proadure worldwide applicable. I hope the risults an I have so broad a Quelified trained person on the specified jobs. the the scients are surved to the ispectation of all concerned bouches public at large.

	*** 2	isitor	BSSK	
DATE	NAME & ADDRESS	CONTACTS	E-MAIL	REMARKS
3/7/2010	Po. Shaikh S.M.			
	New Arsts com & sci	9420241826	smisk of gyerhus. com	very nice tab & excel-
	college shergaon.			working,
3/7/2010	PE. Boeude D.A.			
	New Azt com 2 Sci (dlege	9975906908	Devikazani bozade a gma	Excelent Lab also
	shergaen.			very good waking
3.7.2010	Prof. c.s. Arsule	9423160730	Assule (s Gemeil @.com,	Very nice lab, good
	New, Art, comm. 9 sci collese			Environment, we got very
	shevsan,			Dml. externetion about
				worter analytis. Mahajamsir,
200				pawar mangon & sech pande give
4				erel. Intermation.
CAN ME				
19/07/10	Mayuresh. M. Chakkarwar.	9970401614	mchakkarwar@zediffmai	very nice las 2
AL	ptoject Fellow.		Com	
	Deogini college, Automphad.			Advante instruments.
(mohatikan
	· · · · ·			

	V	isitor's	Book	***
DATE	NAME & ADDRESS	CONTACTS	E-MAIL	REMARKS
21/07/10	Sanjay K. Garhane	9423149701	sgavhane 69@yahoo in	found a complete
	AE (ELM) MIDC (Wreiter weaks)	942314970	-0	good quality labs at
21/4/14	Dalla FT Mashind.	A 60 0 7 14/27		Aurangabad level. Really it is very good thing that
21 17 10	Ravikant T. Marwade DECEBM) MIDC	9772917675		Nicely set up father
			4	all parameters of
L.				water quelts
	AN BE			being tester Mo
~				1 (fai) f

***	** 1	isitor's	BSSK	•
DATE	NAME & ADDRESS	CONTACTS	E-MAIL	REMARKS
218110	P.J. Kangari	9702991666	(emile Ori	Laborating is
	CE, MIDE, Manded		ce @ uide noty ind	
				very good. It is creditable that
				this las has
				received 150 9 NOO
				officers e staff or the lab are
				nice and performing
	Ball Alla		5	very good.
	No. Y / Caller			Bir
prior	, Dr. J. N Bhylet			218/10
16 8/1	L. N. U. Frezen server	105 94229428	45 bhussein eycho.	in Very much impressed
10 m	Harren Ausepeld	1. 201	· · · · · · · · · · · · · · · · · · ·	by office the subes
				up data laboratory officers (
				stepp working is woperative & illughtated the importance of
	A A			water analysis. Hence it is
1				water perpise to save the
	· · · · · · · · · · · · · · · · · · ·		1 26810	life huma leist Animale.

3	1	Visitor's	Book	***
DATE	NAME & ADDRESS	CONTACTS	E-MAIL	REMARKS
١.	Sunit J. Rohaterad,	9822111627	-	Laboratory is very good
	Back Asstt.			2 wome dome very
				nice, staff writen is
				co-obserative, e giriosog
				geidelins,
				Im
0				· · · ·
2,	श्री प. सि. सारकट उप किसाशा में अधिकारी	9422272942	-	प्रभोशका अन्यावर्ग आह.
				त्यागुके सावेस्तर-या-यागानी सोभ हादी आहे. या-प्रयोग
			2	साम साखा कार्ट. मा- प्याग हाप्रेच्य स्वाना उपयोग हो देख
	16 1000 3			थाया अक्तानद् मिल्ला
	i de la companya de l			quer
				J. 3. 181 .

DATE	NAME & ADDRESS	CONTACTS	E-MAIL	REMARKS
₽.12.10	रविन्द्र नापायगतब कुलकार्म "प्रकल्प झणिकारी, भ.पुरी कुक्षी प्रतिथान, ऑरेगानाद.	9423150067	Yavî Kalkarni 1969 @ g	मार्क. (am उत्तापाकी प्रयोग शाखा अलेग उ त्कुहार अलुन मेके कार्य पहली, व कार्या प्रकारी लढभद 'वाव्याकार जोडी आहे आपलग आगा भी कार्याल तारिङ मुजिर्द्ला. (२ 44)
19-3-11	Shālnıglina Parshram Pailiða Executive Engineor, WALMI Aurangæsad			The lab is well equipped with highly sophisticate instruments having reasonable accuracy
				of testing of water quality Zundfan S. p. pailwa

		CONTACTS	E-MAIL	REMARKS
2/3/11	Projakta K Nawale	9272652166	pk. navale agmai	The lab is having
	Dept of Chem Tech		· com	very well instrument
	Dr. BAMU, A'bad.			and water analysis
				facilities. All the
				Dinethods of water
				analysis were resplain
				to us by staff
				members.
	A			Penauale
23 3 11	Quazi Munnaza.	9665519784.		The lade is well
	Rept. of Chem. Tech.		g mail . com .	maintained and
	Qui BAMU, Abad			is well equipped
				with instruments
	and the second s			even the staff
				in their work
				and 20,00 min
				and garee us much knowedge
	a suffer a s			is nevy seend

DATE	NAME & ADDRESS	CONTACTS	E-MAIL	REMARKS
29-03-11-	Siddhaut Patil	9595921272	sidelove @gmail	The eab ?
	- Summaria - Sura	15 15 1-121	com	
				with standard
				norm's.
				Dud we apre
				effort's of staff
				very linech-
6/4/11	cf. A. Shah	9422291141	anikunj-1 @ Lotmail.com	2ts well maintained
'				by equipment and st
				and more so by profe
				Calibre The staff is
	and the second second			vened with industrial
				-professional procedures
	- 3-21 ⁴			The exposure to B. Fe
				third year students . laboratory has proven
- /				unique experience :
				my sincred thanks I
				organisation for its,

DATE	NAME & ADDRESS	CONTACTS	E-MAIL	REMARKS
24/06/11	DR.S.M. Deshpande			I have seen the
	Head, Deptt. d. Geology	9421303658	dr_smd Carediffici	· Lab and the morathy
	Govt. Institute of science			region this lab is
	Aurangebad.			well canined and
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				procedure followed
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	als hat belo	*		lab is good tor
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DATE	NAME & ADDRESS	CONTACTS	E-MAIL	REMARKS
30/6/2011	न्हनी राशवंत कार्नुन कांकुश समुप्रदेशक स्रांत सोहाहास कारोग्य केंद्र मुक्तिताडी कोरेंगालाव	7588696538	Jashwart · Ankush@)· Gmai f	संत रोधिवास झारोप्य केंद्वालील २५८०८ प्रकल्प मेल्लीत झसलेल्या झारोष्य मेल्लीन स्वरुख वयास्य
	BY BYCHIS GIVINITIA			भागाना स्पथ्छ व सुख्य पाणा वालिक फूक स्पष्ट केला तसेच पूर्व्येक रहेनामधोठ भेट रेडन सर्वाना स्वमाजाइन सांग्रीतले ही पाहिती
				कामच्यासाही कालिश्वय मरलापी ताहे या माहिलोग्रहे क्रांस्लीमधील
				लोकारी 80'7. आमार हाराब अफ्रमा- हारावाद-
				Hond 4. 30/6/2011



प्रभावत A/C हॉल जबस्य स्वाधिय के जबस्य स्वीप्रिय प्रचामृत 9403900280

हर गुरुवार चलो सम्राट आटा जिमेनी बनावती तुव 170/- 65/алтина, ентентина 14, виде Пред. «Муната. 9923009366



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इटरियम अग्र (१३ अग्राको) एका विविद्यात आपन्य इत्याचे विली होये प्रकारत अवस्तरीत इत्रेडिएक विज्ञे अन्यूत कोर्ग्स स्टर ओक्स्टले जोग

अतिमार अ प्रतिशिक्त करोत प्रिक्वीशाहरूको ५० देवरे ५० देवर विकारणती पहिला अतिम वाड्रोकरको सर्वित्र वाड्रोकरको उन्हें प्रवृत्त स्वित्र वाल्याको उत्तेत स वार्वे प्रवृत्त सेवित्र वाल्याको उत्तेत स वा इन्हेल्वर प्रवृत् हरूत.

गुलमंडीची कुजबूज

गावाकडचे कवतिक...

ठावाठाडी कार्यतिक... प्रकार प्रकार पेर्ल्य की कर्डाक प्राप्ते. प्रार्थ्या केला केल्ह वरी प्रत्यात्व प्राप्तामध्ये भेड्रेष्ट भों प्रत्यात्व केल्ह का केला कर आते. प्रार्थ्या कर्ड एक देश प्रकार आते. प्रार्थ्या केल्ड प्रकार के प्रकार के प्राप्ता केल्ड पार्था प्रयान केल्हा प्राप्ता केली का प्रकार के प्राप्त के प्राप्ता केली का प्रकार के प्राप्त के प्राप्ता केली का प्रकार कार्याहर केला कार्या प्राप्त करें। प्राप्ता केली का प्राप्त करें। प्राप्ता केली प्राप्त प्राप्त का प्राप्ता केली प्राप्त प्राप्त के प्राप्त प्राप्त केली प्राप्त प्राप्त केली प्राप्त केली प्राप्त प्राप्त केलि प्राप्ता केली प्राप्त प्राप्त केलि केली प्राप्त केली प्राप्त केली प्राप्त केली प्राप्त केलि अग करा झाल, तरिप इसने महत्व्य नडीस्थ आती. 'क्यमन्त्राचे पित कडिंग्ल्यदेवीपा आले अठे: स्पूर्वन्त्र प्रतित्वनार्थं आठे पार्थ होत अठे: क्य, बन डाडिल्व महत्व्य प्रता यहा कार्ड: व्याहल्य 'कडेंग् वेत्र आहे. पुलबंडोग्वे कुतब्दूत संघर्त

प्रश्न वाचकांचे

रोते त्या'क प्रत्न वार्वाने व वाराजने वार्वान विषय उदयोग करनागर रूप वार्वान विषय उदयोग करनागर रूप उदयोग कर्मा के प्रतिकार वार्यान के प्रते के की परिवार विषय वार्यान के का किसा वार्यान का किसा वार्यान के प्रति प्रतार वार्यात का किसा किसान राज्य वर्ग-क्रांत्र किसान राज्य वर्ग-कार्यात विषय प्रत्नकार्यात

गवायत्र प्रस्तवरात संबोधायत् प्रत्य विग्रोत राष्ट्रला गालित विग्राज प्रत्यांत प्रसिद्धी दिले गार्डला अगले प्रान makarand k@mh.bhaskamat.com या ई-येलचा बिटवा १८९०३२९६७७ या बोबाइल्बा एनएबएबहुने प्रान्थलेल.



१२ पैकी होतात केवळ ३ चाचण्या, प्रयोगशाळा नसल्याने आरोग्य खात्यावर मढार...

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कोणत्या आहेत आवश्यक चाचण्या? 🔉 पीएच टेस्ट

» टेम्परेचर टेस्ट » डिझॉल्ड ऑक्सिजन टेस्ट » ইমিরহ্যু अल

क्लोरिन टेस्ट » टबिडिटी टेस्ट » क्लोराइड टेस्ट)) नायट्रेट टेस्ट » फ्लोराइड टेस्ट

🔉 फॉस्फरस टेस्ट » आयर्न टेस्ट » हार्डवेस टेस्ट

» अमोनिया टेस्ट यासोबत क्वलिफॉर्म बॅक्टेरिय ततेच बेंधात टेस्टचा रामवेश आहे.

घरातील नळापर्यंत येणारे पाणी शुद्ध, स्वच्छ आणि निर्जनुक असावे यासाठी पाण्यावर किमान १२ प्रकारच्या तपासण्या होणे आवज्यक भारत मात्र, महानगरपा लिका यापैको केवळ इ ते आहे. मात्र, महानगरपालिका यापैको केवळ इ ते ६ चाचण्या करत असल्याने शहरवासीयांचे आरोग्य योक्यात आले आहे. याबाबतु केंद्र अगाग्य धावचात आए आह. यावावत कद्र ज्ञासनाचे कडक निर्देश असतानाही औरंगाबाद मनपाकडून पाण्यावर आवश्यक त्या चाचण्या केल्या जात् नाहीत. काही ठरा्वीक चाचण्या होतात त्याही अद्यापावत नसलेत्या आरोग्य खात्याच्या प्रयोगशाळेत. अन्य महत्त्याच्या बाखण्यांसाठी नाशिकला धाव घेतली जात आहे.

मंत्र अर्थः १८२३४४४४४३४ – पत्ररोळ्याला काय होते?

जायकरवाडी औरंतबबर सहदारा वार्वपुरावरा करपाऱ्या अवरवत्वी प्रत्यात्री तेऱ्या सार्व विरुवांचपूर बुदेश प्रत्यावे उत्तरत सत्राजे. सार्वी विरुवांचुंत रत प्रदारत हिरवा रंग्ये तात्रा अन्य क्रांत्र अन्य त्रांत्रे. तांत्र प्रत्याते घरणगांचे पाणी पाण्ये बहुवे त्याव्ये क्रिके विवर्कत अन्ये केंग्रे, व्युवी स्त्रेत अत्र प्रदेश किर्य प्रदेश कर्या प्रदेश प्रत्यावे सुहारों पाण्ये बहुवे त्याव्यावर्थी यांत्रा उपत्रस्य बत्या हे बहुवे सविष्ठता प्रवेशस्वरंत प्रत्याज्या आहे.

STATISTICS.

अग्रयावत प्रयोगशाला प्रस्तावित आहेत

QUECTION 30001
के सार्वाध्य प्रारंग प्राप्तारा ३४० ते ३०० प्राप्तार पर्यं के किंव सालत रुप, या सर्वेय प्राप्तार पर्यं के किंव सालत रुप, या सर्वेय प्राप्तार पर्यं के किंव सालत रुप, या सर्वेय प्राप्तार पर्यं के तिराप्तार प्राप्तार के तथा के प्राप्तार प्राप्ता प्राप्तार प्राप्तार प्राप्ता प्राप्तार प्राप्ता प्राप्तार प्राप्तार प्राप्तार प्राप्तार प्राप्ता प्राप्तार प्राप्ता प्राप्तार प्राप्तार प्राप्ता प्रा प्राप्ता प्राप्ता प हास सारवारी, करा जीवांच कर

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आपकेस आगाप भाषात्मवाग हिन्द कामवापूर गोवाप्य अवार्या काम हुआवादी आसार्गाचेस विद्युप्रेवणाचीन पुरस् आता आपके देवियों प्रया पार्च प्रयानायां क्यें किस आदी देवियों प्रया पार्च प्रयानायां क्यें किस जिसे आपने प्रवार के आताप्य आती.

डिझाईनर सारीज् संपुर्ण फ्रेश स्टॉकपर

डिस्काऊट धमाका

जोरोंचे शुरु है । १ चुडही दिन और..... २ दर्धर, बेवरूस १८ वेह के अस, मास: ब्राजस, ÷ ---

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रायंच कोणिंग सेराइंडाले एक्टरेएससीच्यावंत पूर्णपेळ संचालक नेवले जावार आहेत. बाह, ही प्रक्रिया पेजवाज राधानक नेपार्ग जायार जांत र भार, भार जांत आहे. व्यक्तित दिवरण वापूर निकास तालेपची सेता आतावने काले दिवर सात्री प्रचेत देवरण प्रांत्य उत्पन्न आपि अनुभावा प्रचया में बेदलारा निकास ता सेता बदस प्रदूर पेडाल. हे. अभिकारी मन सातून तिकास्त्रेल. पावाकर कोर्प प्रचेत. हे. अभिकारी मन सातून तिकास्त्रेल. पावाकर कोर्प प्रचेत. आतं जा त्यांचे प्रधानन का राजेल टोपे, राज र रह





Photographs